

Implications for Exploration With The Use of High Resolution ICP-MS Technology

Eric L. Hoffman, Yakov Kapusta and M. Dzierzgowska



The toxicity of trace metals in the environment is an area of increasing concern. Toxicity of the trace metals is highly dependent on the molecular species to which they are chemically bound. For example Arsenic 5 is very toxic (check) whereas Arsenic 3 and the organoforms of arsenic are much less toxic.

Some of the species of interest for Arsenic is shown in Figure 2 .

Introduction:

- **What is HR-ICP/MS?**
- **Advantages and Disadvantages**
- **Applications:**
 - * **Hydrogeochemistry (Au+PGE)**
 - * **Biogeochemistry (Au+PGE)**
 - * **Lithogeochemistry (REE)**
 - * **Pb Isotopes**
 - * **Metal Speciation**
- **Conclusions**



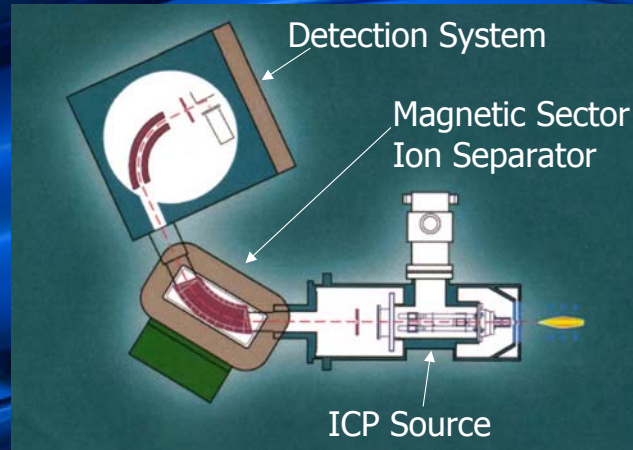
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**What is the difference between
High Resolution ICP-MS
(HR-ICP/MS) and Quadrupole
ICP-MS Technology?**



High Resolution ICP-MS



Advantages of HR/ICP-MS vs. Quadrupole ICP-MS

Resolution – to separate interferences

Detection Limit – to detect metals at their natural levels which are typically very low



The first category is simple redox states: the species is a well-defined cation or anion. The second class is organometallic compounds. There is another group for some metals which includes metal complexes: the metal of concern form a coordination bond binding to a ligand.

Monitoring elemental species requires analytical technology with sufficient selectivity and sensitivity to resolve and quantitate the individual species at ultra trace levels, since an individual toxic species may constitute only a small fraction of an element's total concentration in a sample.

Modern HIGH PRESSURE LIQUID CHROMATOGRAPHY (HPLC) techniques are fairly versatile owing to the diversity of separation technologies available (multiple columns).

The separation mechanism complementary to HPLC is the capillary electrophoresis technique.

**High Resolution ICP-MS vs.
Quadrupole ICP-MS (Conventional)
Limits of Detection**

Element	HR-ICP-MS	ICP-MS
As	3 ng/L	30 ng/L
Se	20 ng/L	200 ng/L
Au	0.05 ng/L	5 ng/L
Pt	0.5 ng/L	10 ng/L



The combination of CE with HR-ICP-MS promises to be a very powerful tool for metal speciation. To produce commercial speciation data we require online analysis of CE coupled with the HR-ICP/MS. Use of conventional ICP/MS does not have the required detection limits.

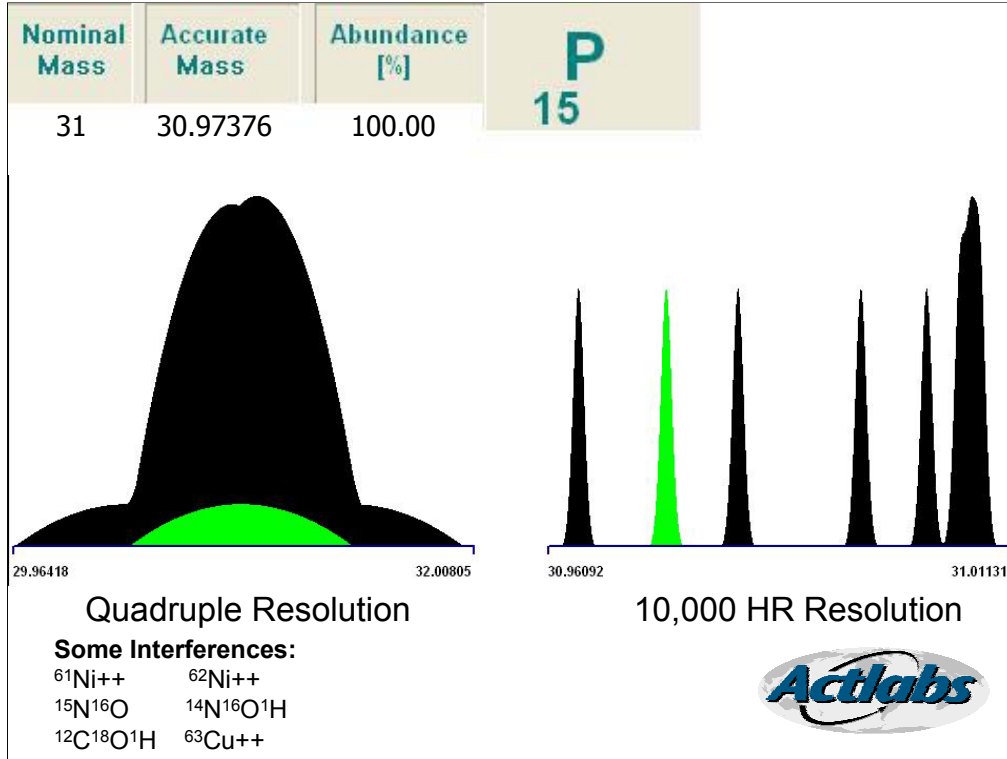


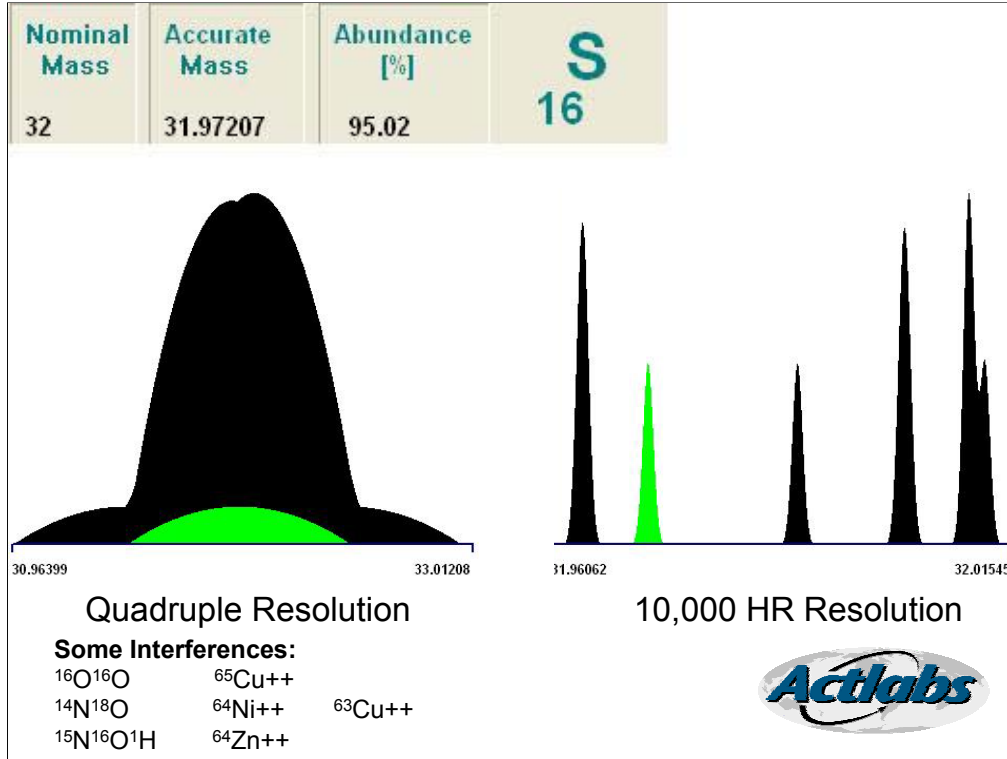
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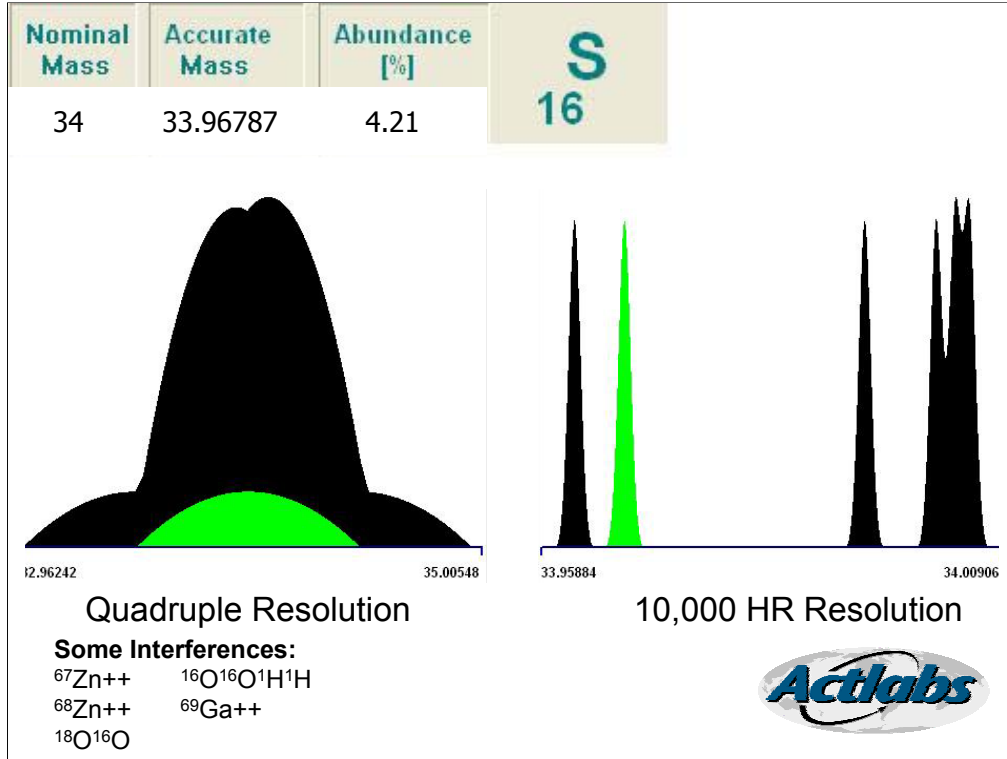
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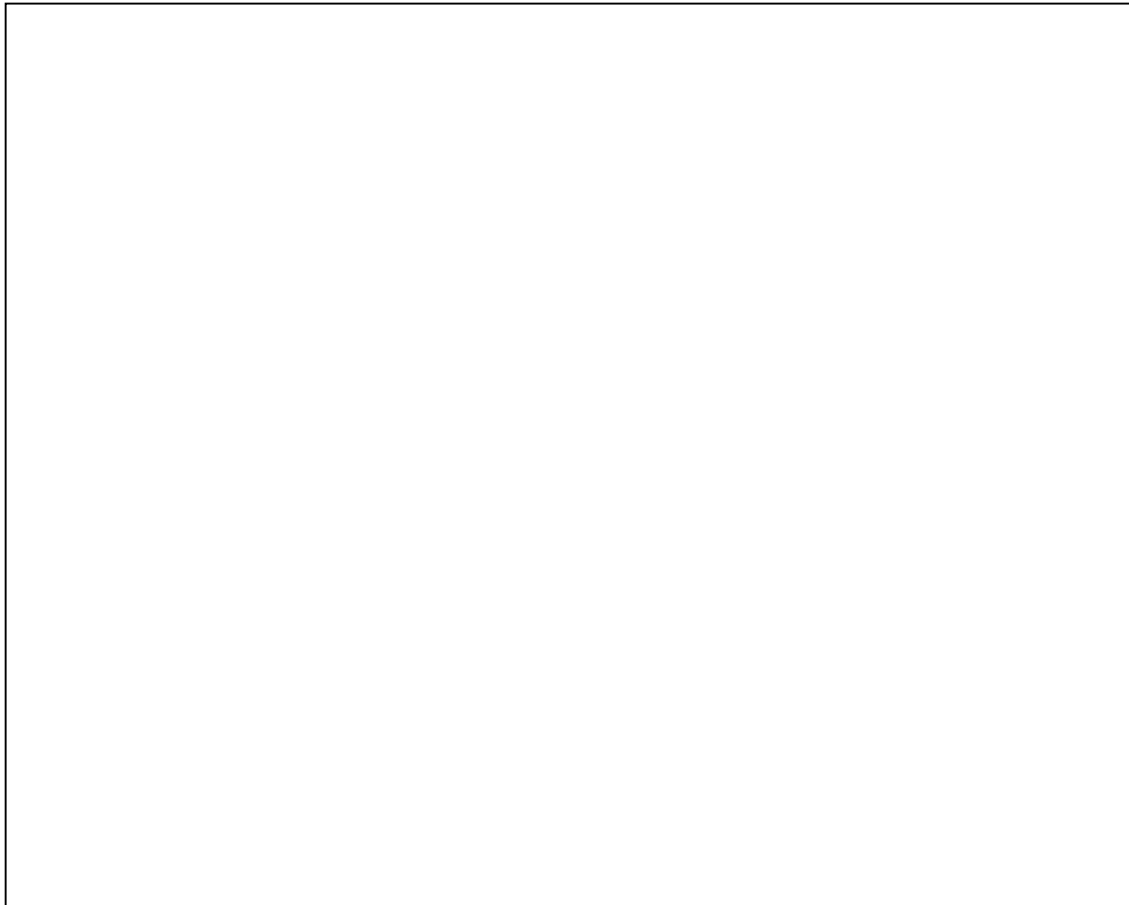
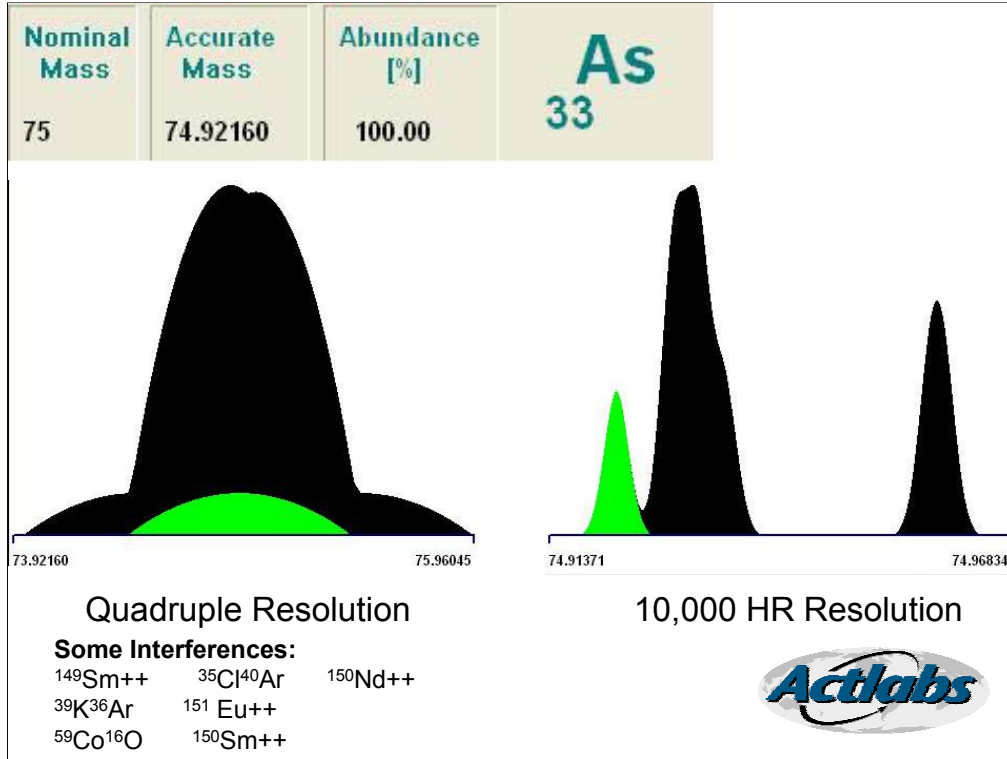
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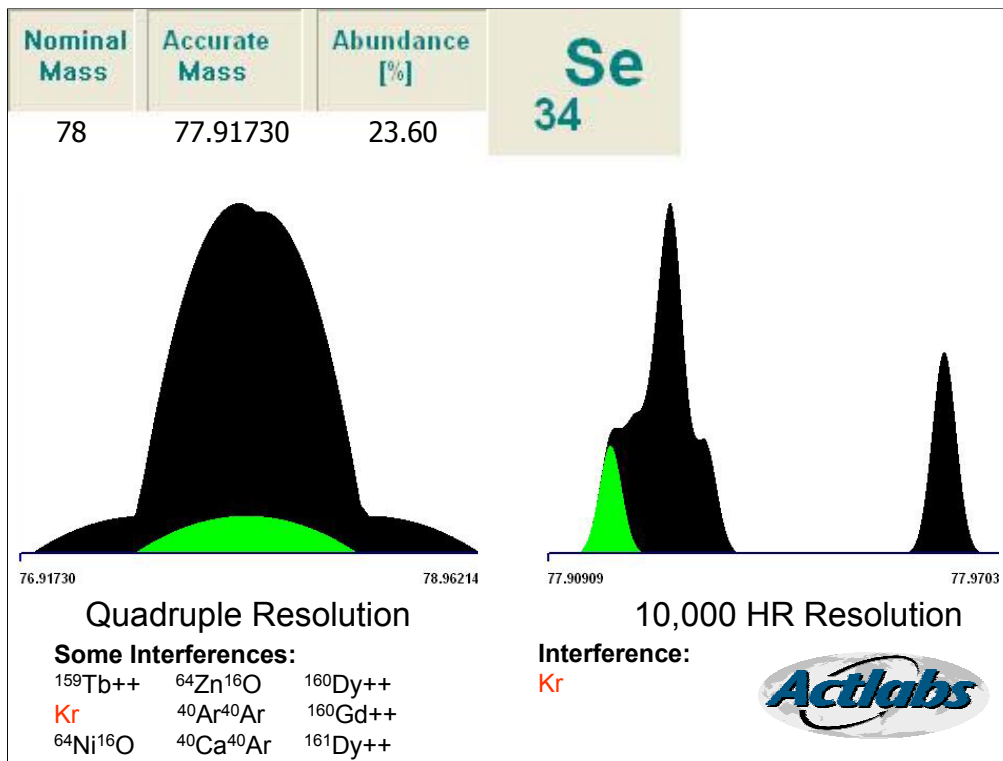
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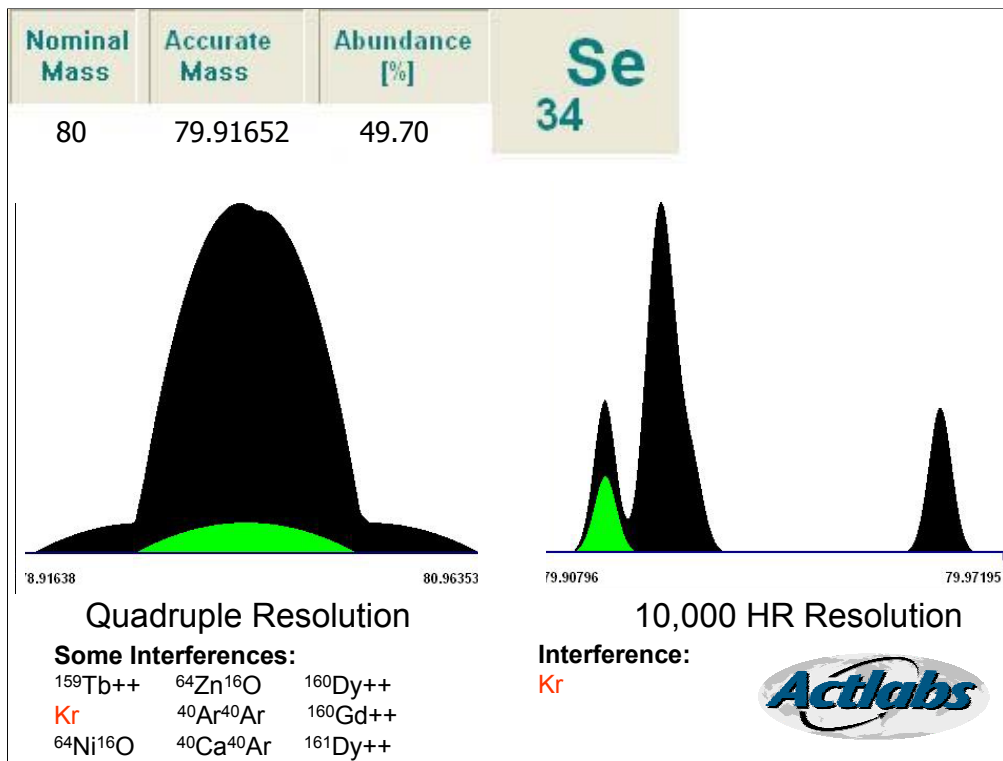
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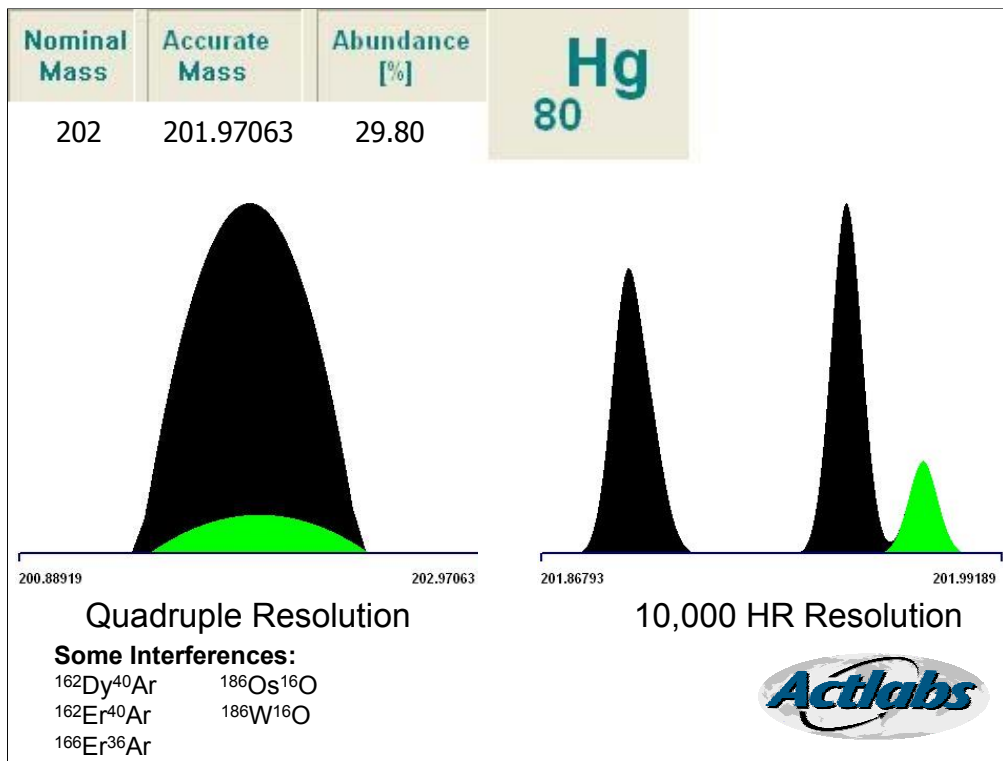
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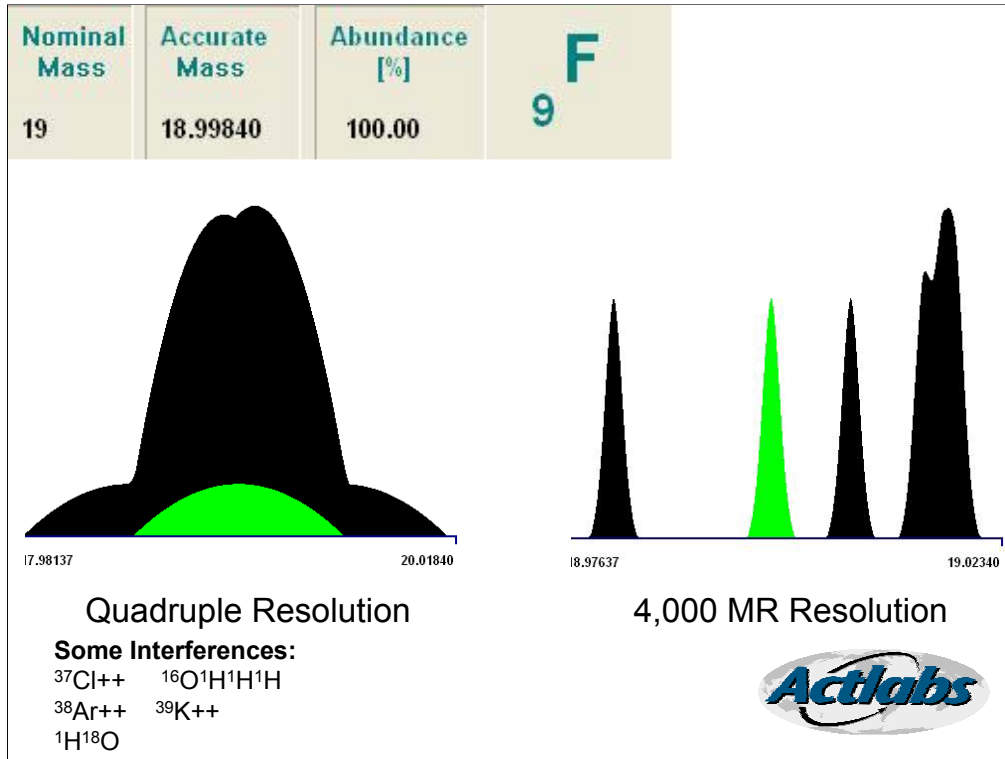
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New Elements Can Be Done By HR-ICP/MS

- Fluorine



Speciation can be done on various sample media.



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Rare Earth Elements

- Barite rich samples have always created an analytical problem with BaO interference on Eu



Speciation can be done on various sample media.

Methods for the Analysis of Gold in Water

- Carbon sachet – INAA (CSIRO Method)
- Direct ICP-MS
- Evaporation ICP-MS
- Evaporation in baby bottle liners - INAA



Speciation can be done on various sample media.

Carbon Sachet - INAA

Method:

- Carbon sachet is placed in a 1 L bottle of water and it is assumed that the carbon sachet will adsorb all of the gold.

Problem:

- Our experiments indicate that most of the gold is adsorbed by the carbon sachet mesh with only ~40% of the gold being adsorbed on the activated charcoal. Gold is also adsorbed on the walls of the polyethylene bottle
- INAA cannot determine PGE at this level instrumentally at a reasonable cost



Speciation can be done on various sample media.

ICP-MS Analysis

Method:

- Gold is analysed directly by ICP-MS or by preconcentration ICP-MS.

Problems:

- Direct ICP-MS does not give enough sensitivity
Natural levels of PGE and Au are below detection limits
- Preconcentration is prone to contamination
- Problem of adsorption on bottles is present



Speciation can be done on various sample media.

Evaporation and INAA

Method:

- 100 mL of water is collected and evaporated in baby bottle liners and analysed by INAA

Problems:

- Slow process and baby bottle liner blank can vary
- Cannot determine PGE by this method



Speciation can be done on various sample media.

Direct Analysis and HR-ICP/MS

Method:

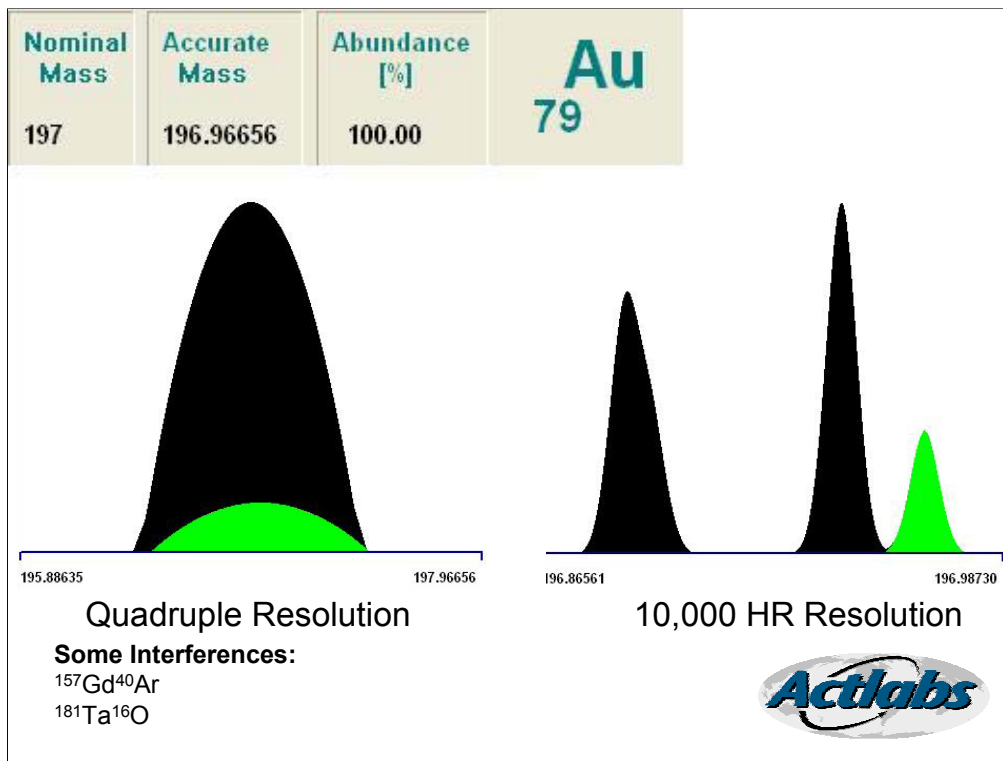
- Samples are collected in polyethylene bottles without preservation required.
- Acidification and complexing agent is added in the laboratory under tightly controlled conditions
- Analysis performed by HR-ICP/MS

Problems:

- PGE background may still require preconcentration depending on local geology



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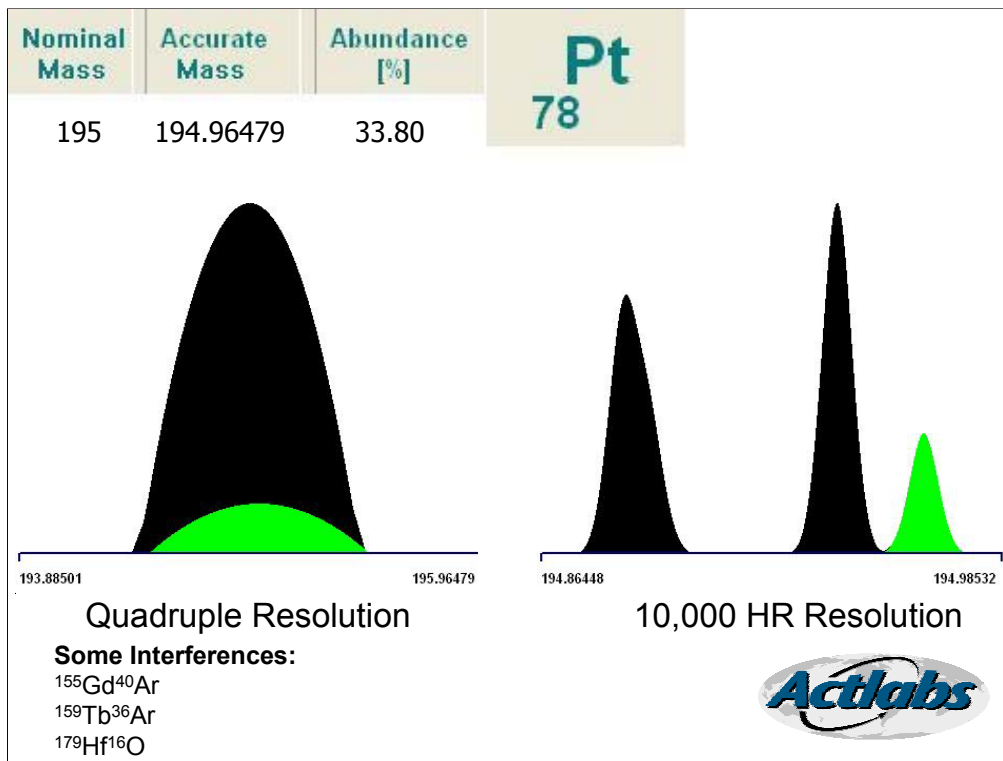


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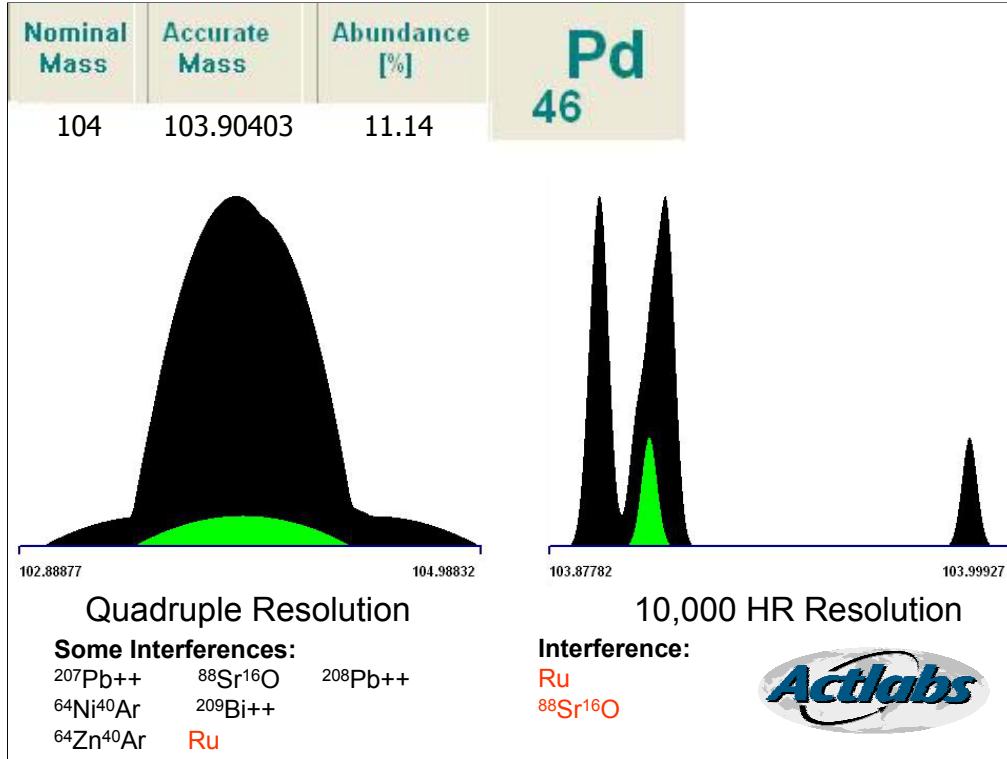


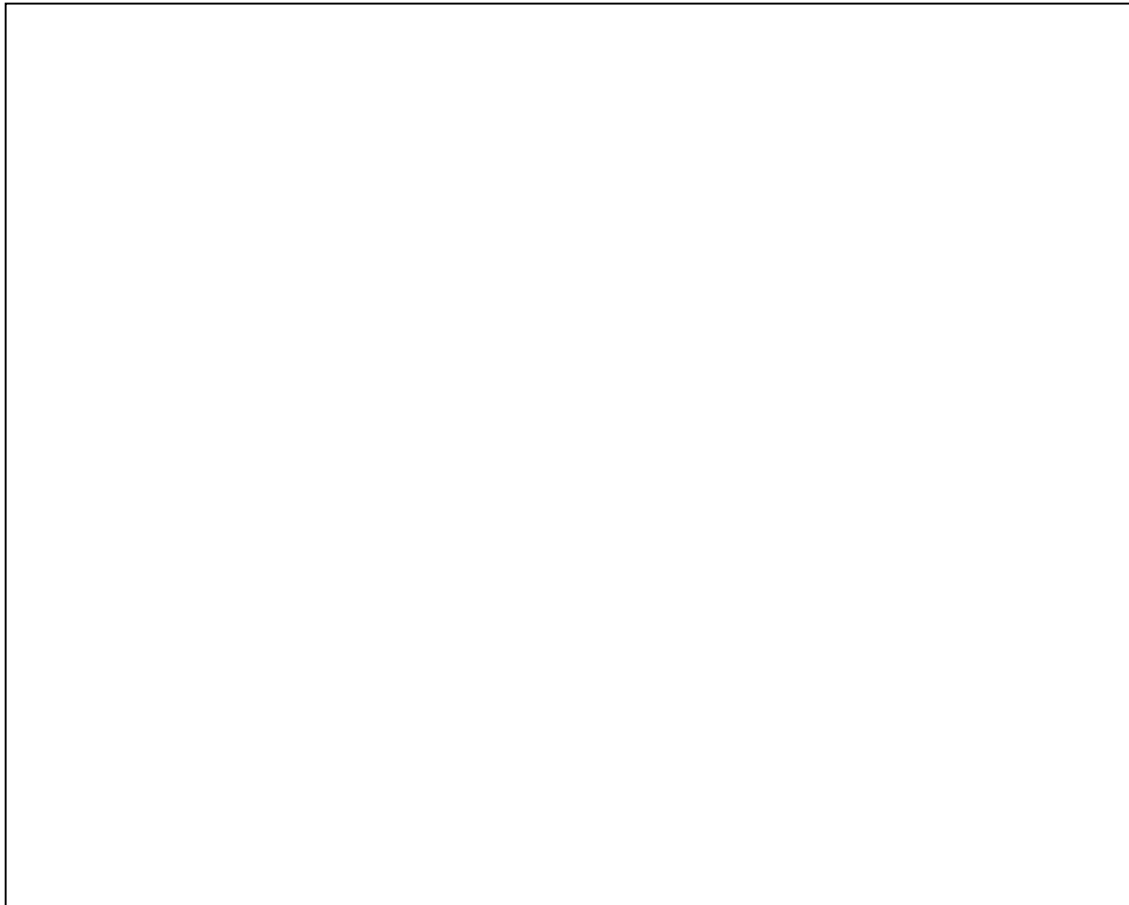
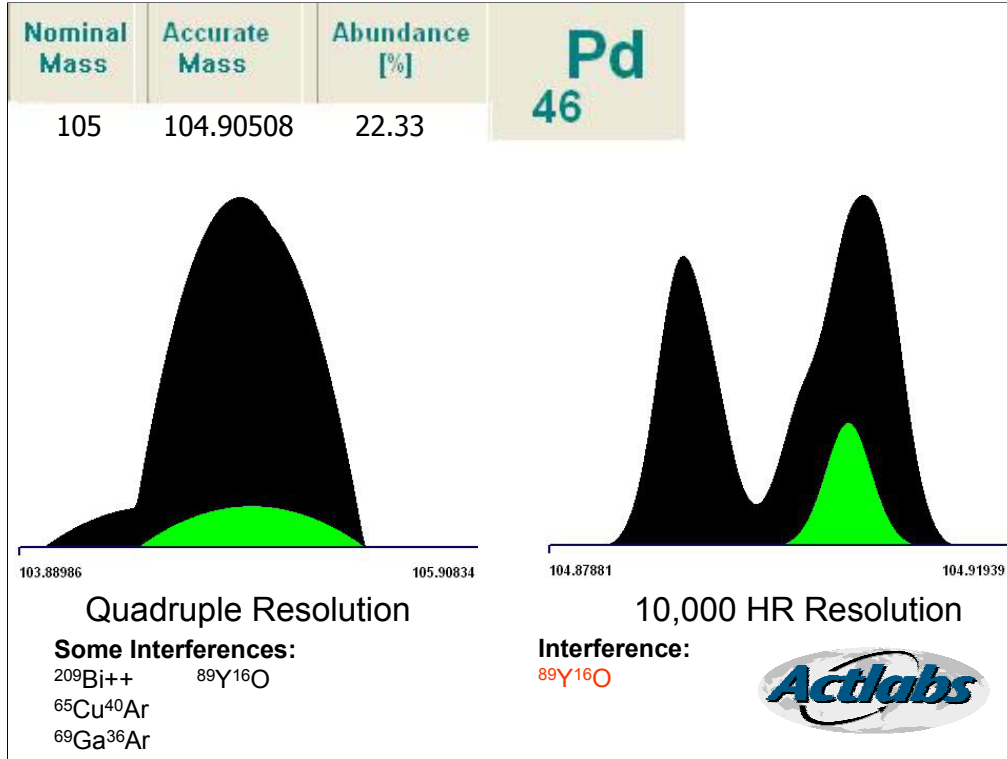
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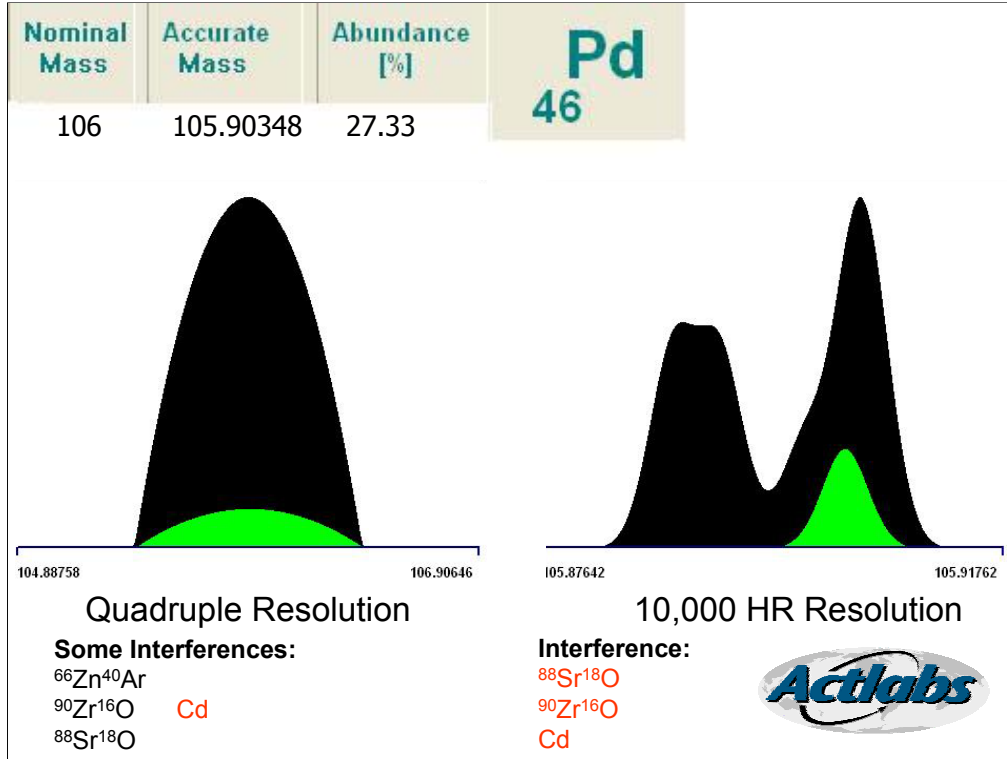
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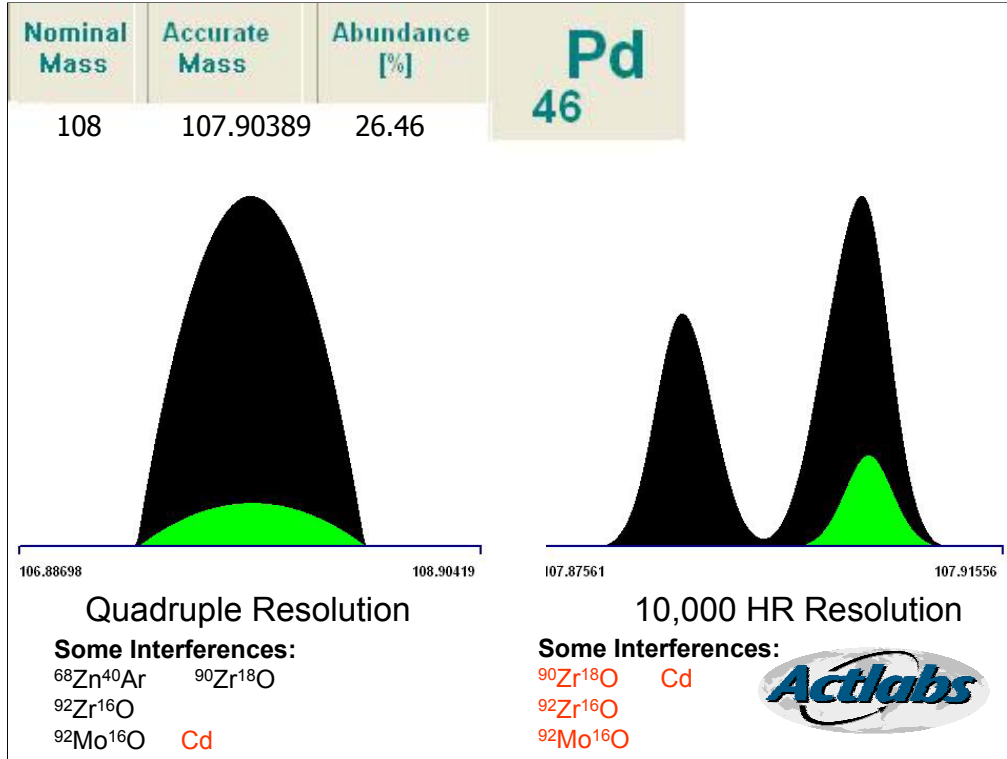
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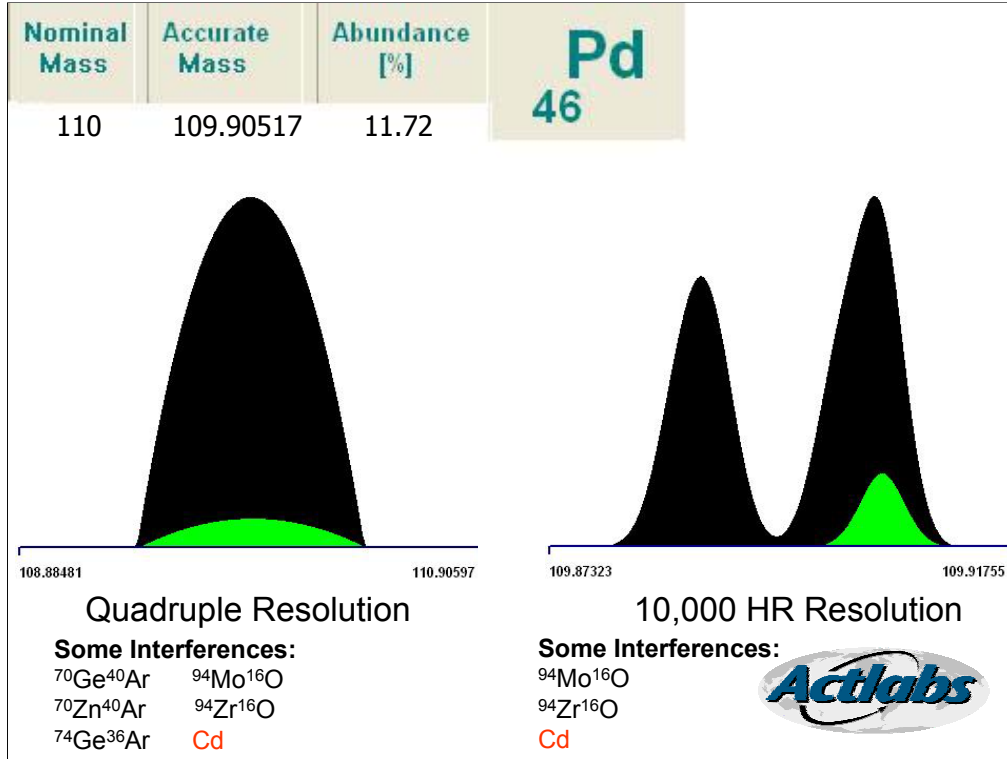
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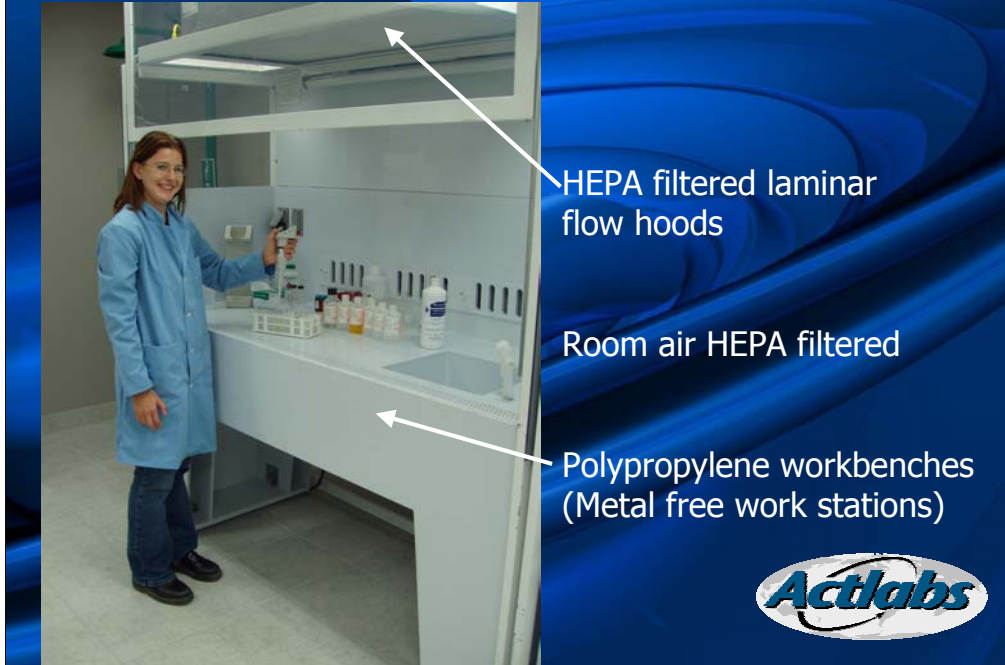








Clean Sample Preparation Areas



How do we do it?

HEPA Filtered Air in ICP-MS Laboratory



How do we do it?

Platinum Group Elements and Gold in Water

- Detection limits in parts per trillion (ng/L)

	ICP-MS	HR-ICP/MS
Au	5	0.05
Pt	10	0.5
Pd	20 varies	1 varies (Cd, Sr, Zr, Mo, Y interferes)
Rh	20	0.5
Ru	10	1
Ir	5	1



Speciation can be done on various sample media.

Biogeochemistry

- Plant material can be analysed directly by HR-ICP/MS for PGE+Au plus many other analytes
- Losses resulting from ashing are eliminated through direct analysis of macerated material



- Background levels can now be reached for most metals



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**HR-ICP/MS Biogeochemistry
Selected Detection Limits on
Dried Vegetation in ppb**

Au 0.01	Ag 1	Pt 0.1	Pd 0.2
Cu 15	Mo 1	Co 0.5	Re 0.1
Ni 100	Pb 6	Ru 0.5	Sb 0.2
As 5	Bi 1	Te 1	Hg 5



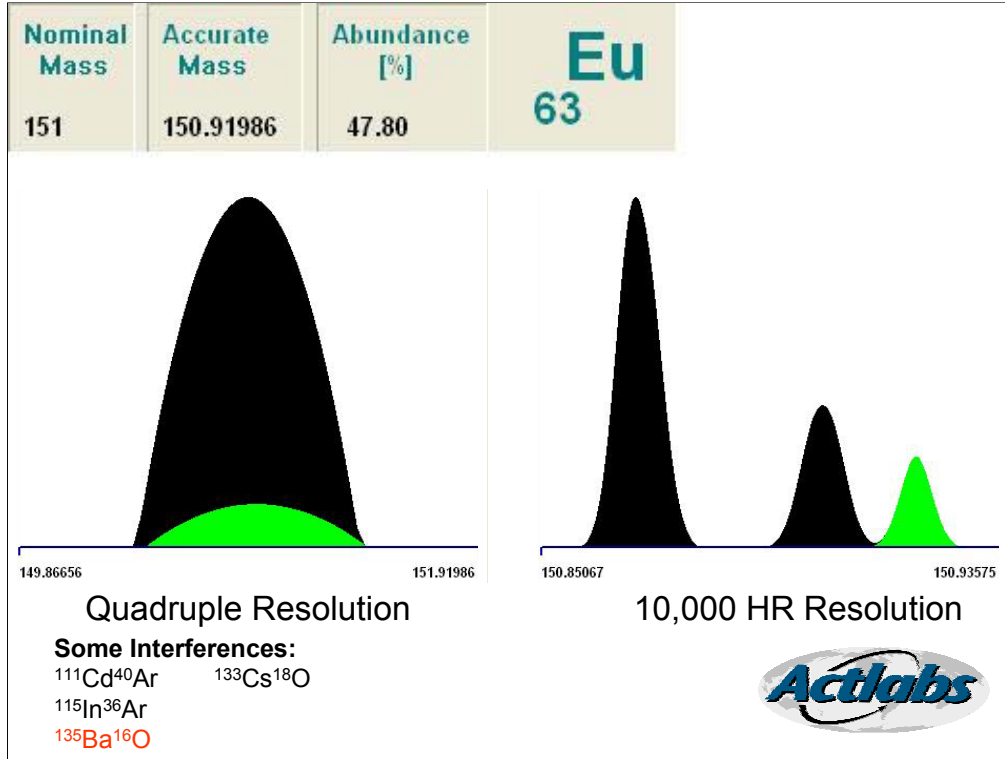
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Rare Earth Elements

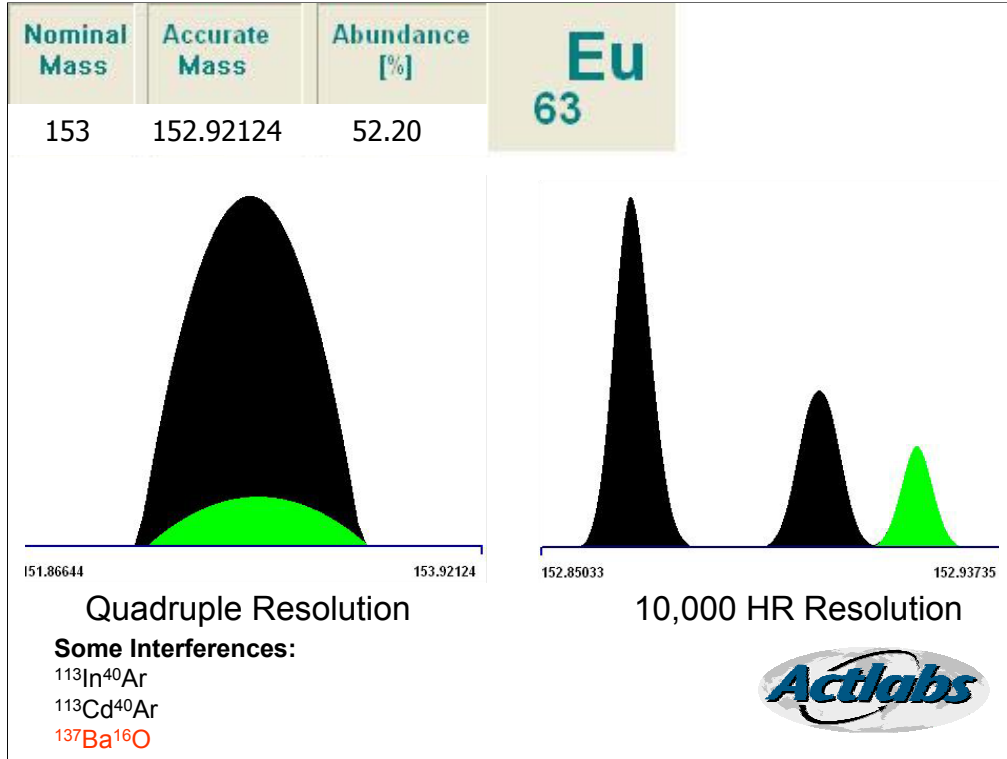
- Some rock types such as very low REE ultramafics could not have the REE determined by quadrupole ICP-MS due to detection limit problems
- Barium interferes on some REE in high barite samples



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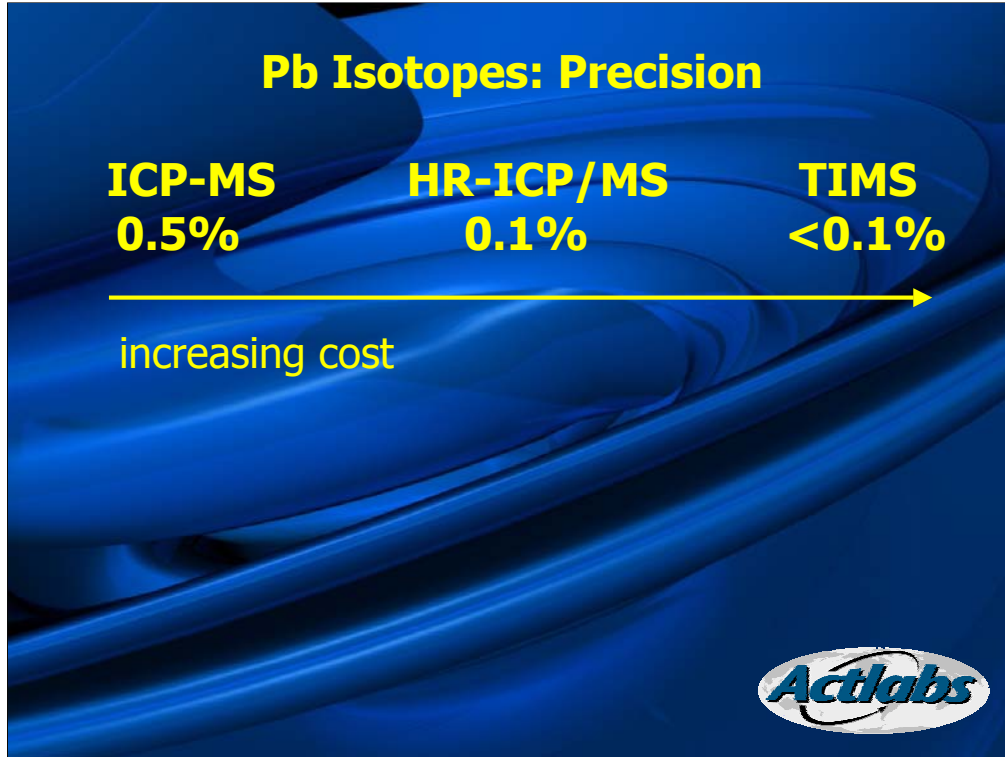
Rare Earth Elements: Detection Limits by Lithium Metaborate/Tetraborate Fusion HR-ICP/MS

Analyte	ICP-MS	HR-ICP/MS	Analyte	ICP-MS	HR-ICP/MS
La	50	5	Tb	10	1
Ce	50	5	Dy	10	1
Pr	10	1	Ho	10	1
Nd	50	5	Er	10	1
Sm	10	1	Tm	5	0.5
Eu	5	0.5	Yb	10	1
Gd	10	1	Lu	2	0.2

Detection limits in ppb



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Pb Isotopes: Exploration Potential

(Modified from K. Fletcher, 2003 GAC Abstract)

Comparison of Pb-isotope ratios for the Swim deposit to ratios for the anomalies might:

- (i) corroborate the relation between them
- (ii) allow the ability of different extractions to preferentially "see" Pb from the deposit.
- (i) TIMS anomaly and HR-ICP/MS anomalies showed similar patterns
- (iv) relatively rapid, inexpensive Pb isotopic fingerprinting of geochemical anomalies may become possible with HR-ICP/MS



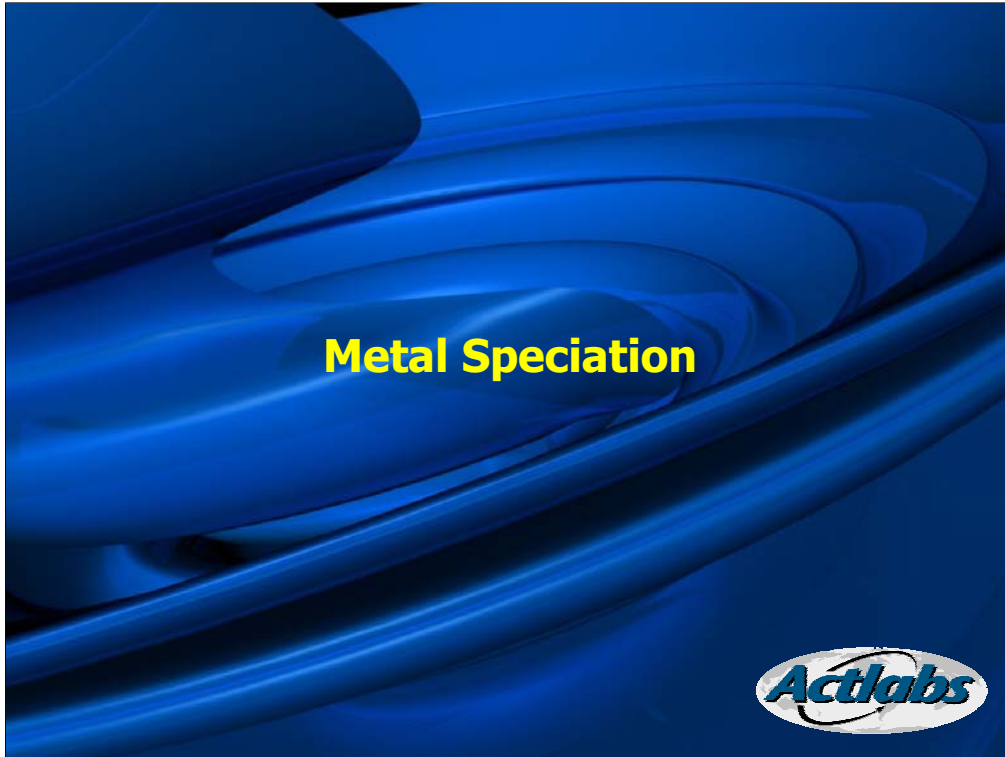
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Example: Arsenic Speciation

As(III) - arsenite

As(V) - arsenate

MMA – monomethylarsonic acid

DMA – dimethylarsinic acid

Arsenosugars, etc.

↑
Increasing toxicity



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Methods for measuring metal speciation?

Chromatography
(for separation)

+

Spectrometry
(for detection)

Examples: HPLC (high pressure liquid chromatography + UV or MS detection)

GC (gas chromatography + UV or MS detection)

CE (capillary electrophoresis + UV or MS detection)

...plus many more methods!



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**How are we going to measure
metal speciation?**

**Capillary Electrophoresis with High
Resolution ICP-MS detection
(CE-HR/ICP-MS)**



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How do we choose the appropriate method?

- ⌚ High resolution on separation –
CE promises the best resolution
- ⌚ Natural systems contain very low levels of metals –
HR ICP-MS provides the lowest detection limits
- ⌚ Geological Applications may have very small sample sizes -
CE-HR/ICP-MS can analyse samples in the μL range



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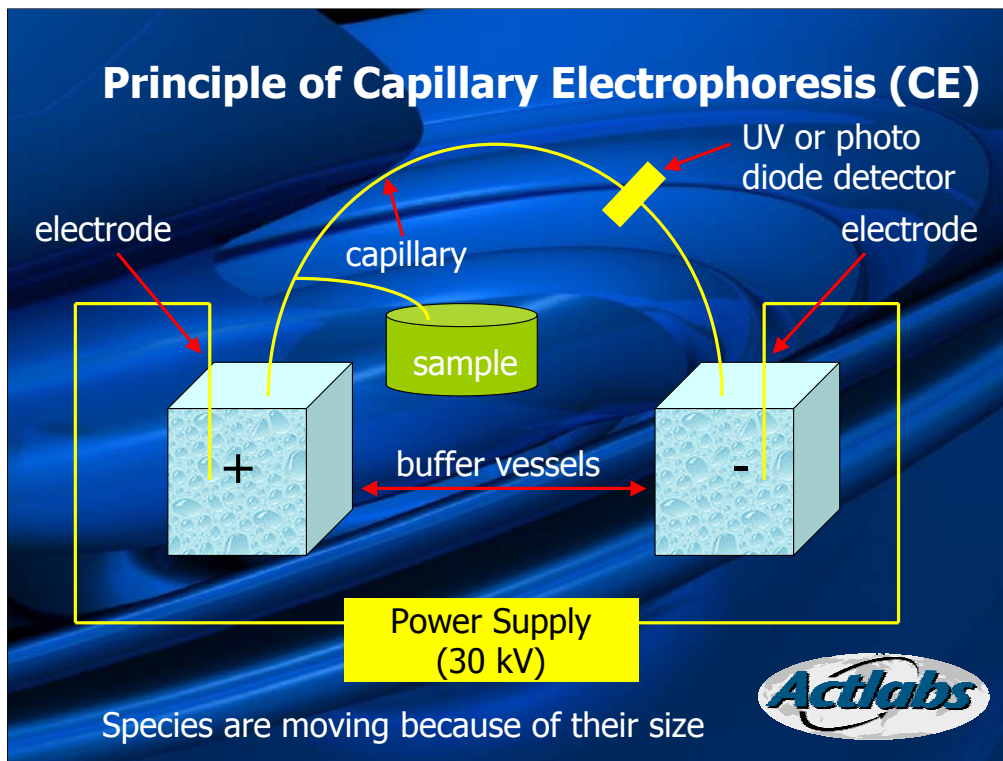


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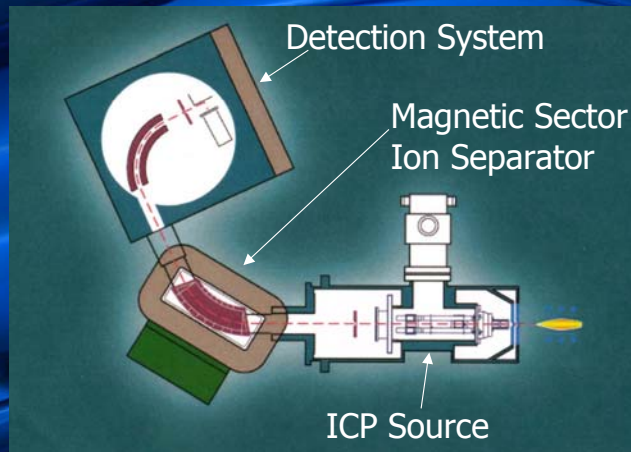
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The high-resolution potential of capillary electrophoresis makes the CE technique valuable for separation of metal species, both organic and inorganic.

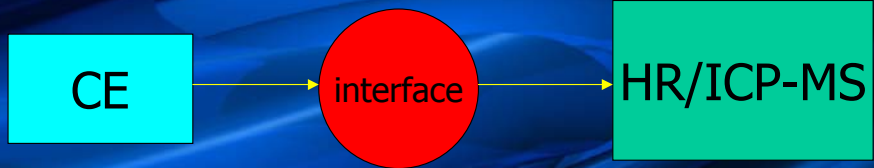
Capillary-zone electrophoresis has a widely demonstrated potential for the separation of As and other species like Se. This includes not only the inorganic species but also organometallic species and species like metallothioneins which are very important biologically. Applications to real samples however have been rare because the UV detection used was often not capable of either detecting the analyte in a real sample or of confirming its identity. The major problem in the application of capillary electrophoresis-based coupling to real-world samples is the small sampling volume which may be available sometimes only a few microliters. This small volume requires a very sensitive detection system to match the naturally occurring analyte concentration levels.

Principle of High Resolution Magnetic Sector ICP-MS



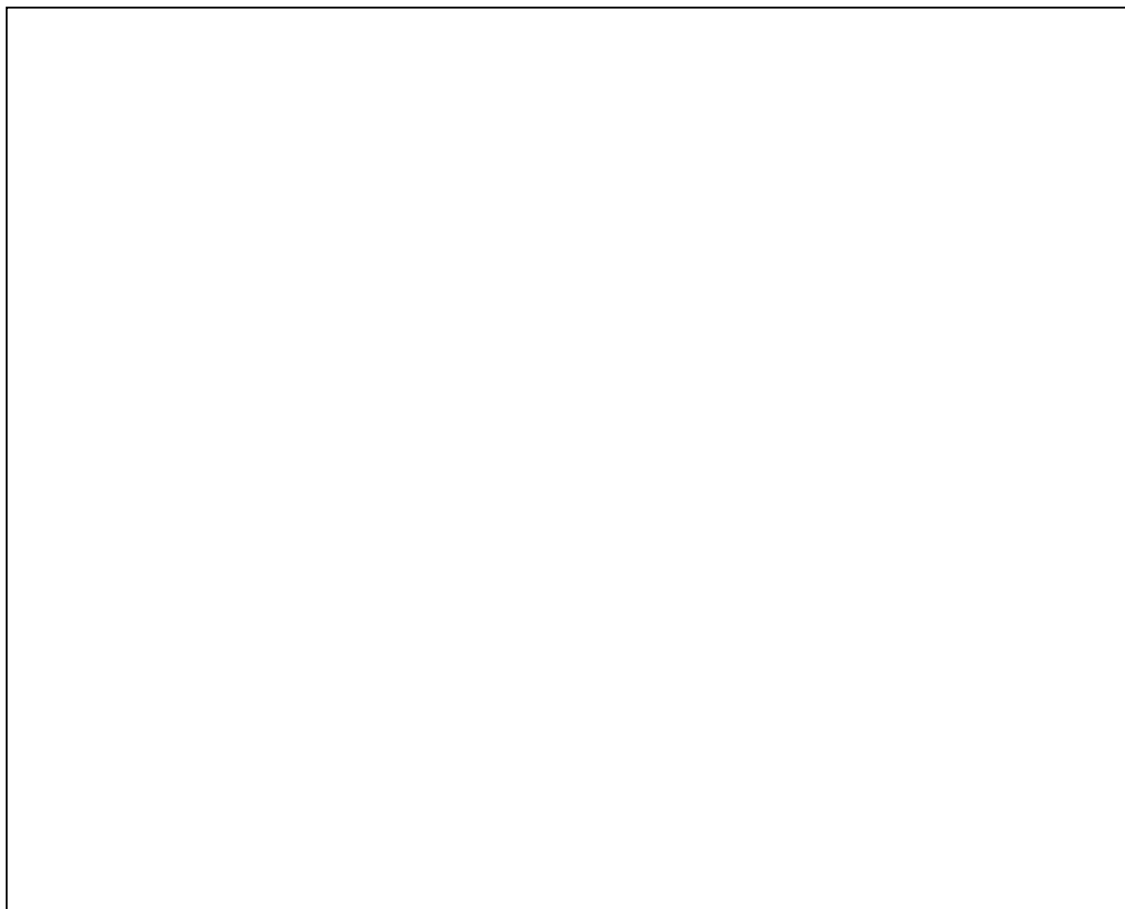
HIGH RESOLUTION ICP/MS (HR- ICP-MS) is an element-specific multi-element detector, **(figure 4 diagram of Element)** providing extremely low detection limits. The sensitivity is 10 to 100 times better than conventional ICP/MS which uses a quadrupole detector whereas the high resolution machine uses a magnetic sector detector.

Coupling CE to HR/ICP-MS

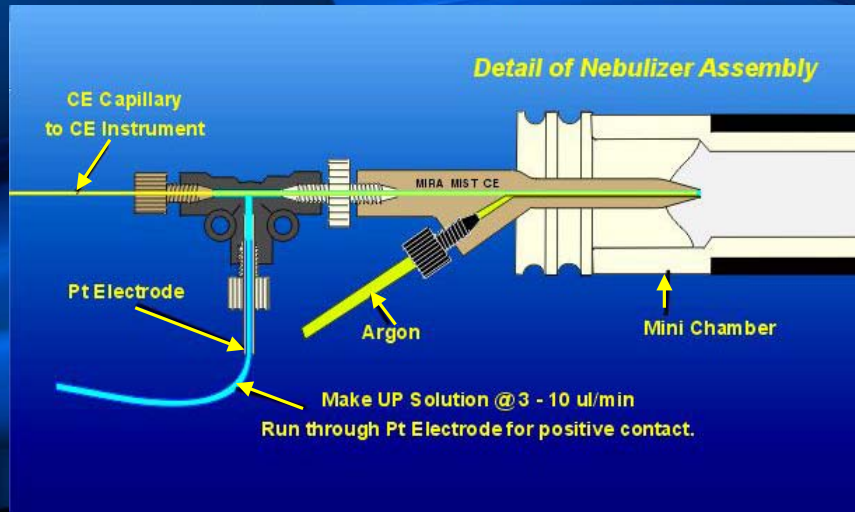


How do we do it?

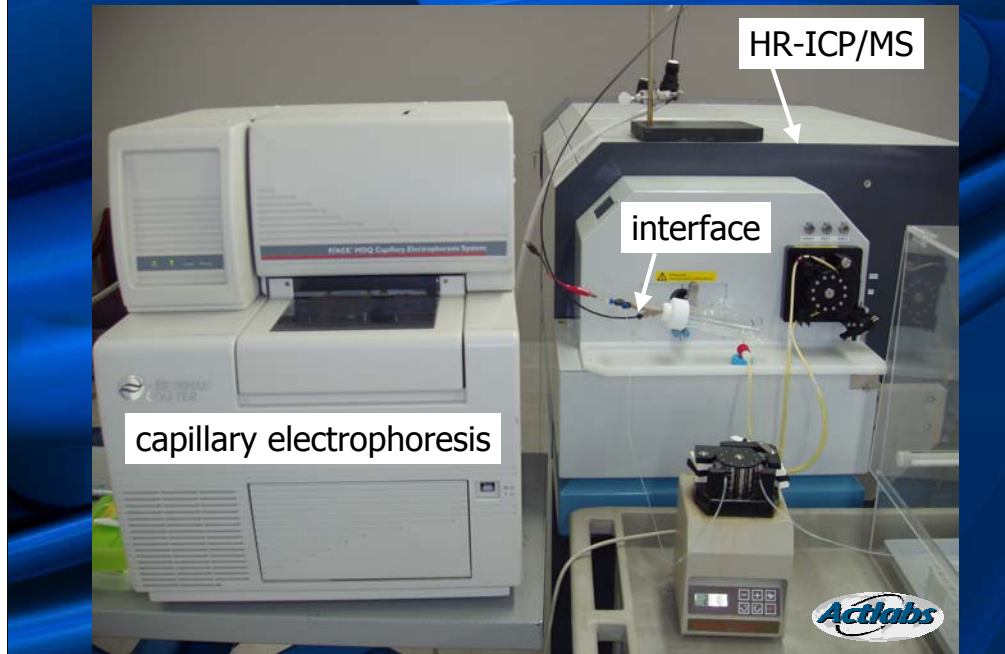
Principle of MIRA Mist CE Nebuliser



CE-HR/ICP-MS Interface

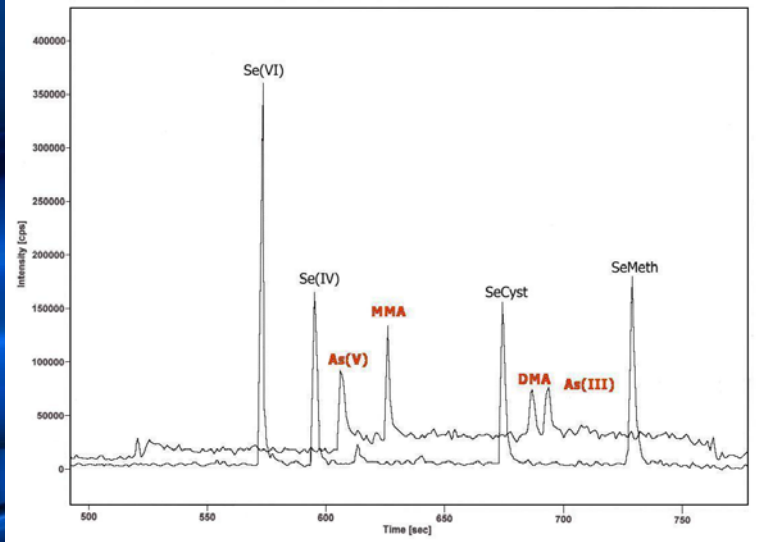


Capillary Electrophoresis and Interface



How do we do it?

CE-HR/ICP-MS is a multi-element speciation technique



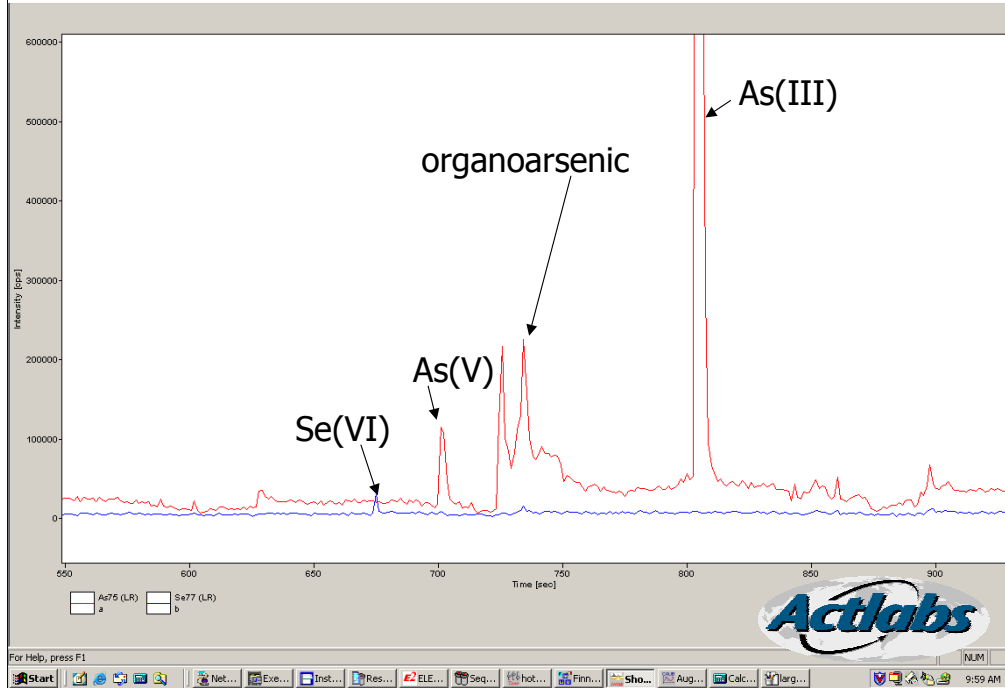
The next 3 slides show As speciation, Se speciation and then the multi-element capabilities by recording both together.

Pore Water Squeezing Apparatus (courtesy of Dr. Dave Blowes, Waterloo)



The amount of pore fluid varied from 2-15mls however a very significant number of tests including Ph, Eh, alkalinity, trace metals by ICP/OES and ICP/MS as well as speciation for As and Fe required that our methodology use only a minimum of sample.

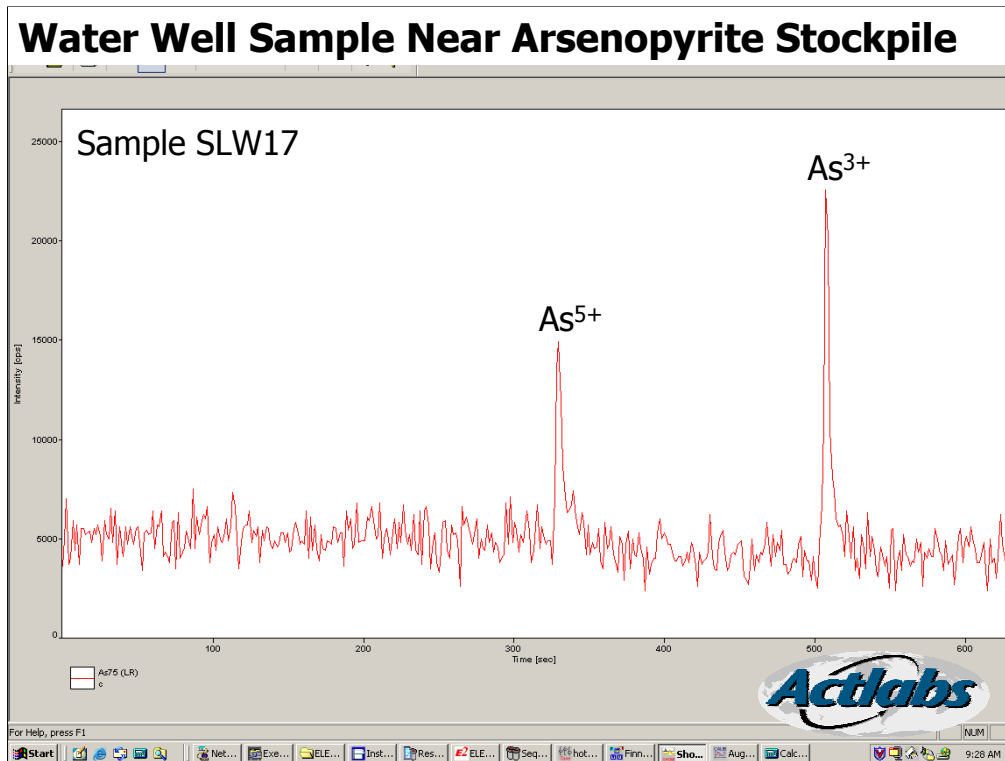
Pore water from mine tailings



This chromatogram shows the As species separated from pore water extracted indicating that most As was As 3+ although As 5 was also identified as well as 2 of the organoarsenic species.

We have also applied this technology to the analysis of stream waters for routine environmental monitoring and are applying the technology up the food chain to biological species.

This technology also promises to have application in mineral exploration where we might expect various species to be present in the vicinity of orebodies that are not generally available in unmineralized areas.



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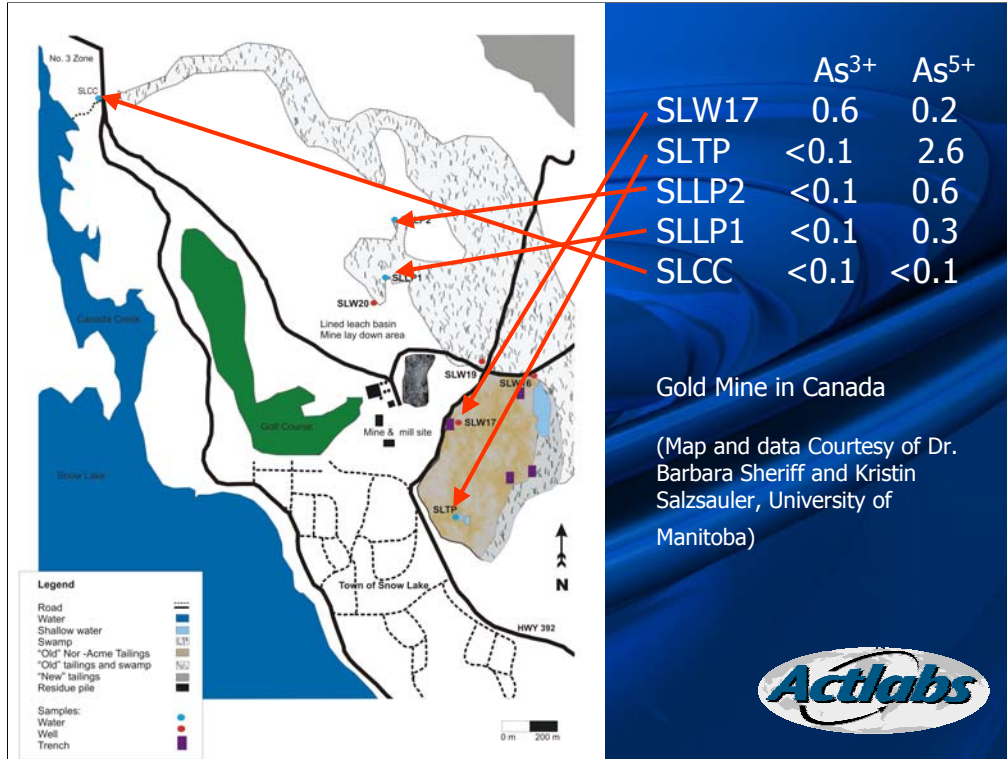
Metal Speciation Application to Exploration?

- **Mapping redox potential**
- **Unique metal species associated with ore deposits? Can we use these unique species to differentiate good anomalies from bad?**

Linda Bloom in 3D Geochemistry CAMIRO Project is proposing to test this hypothesis



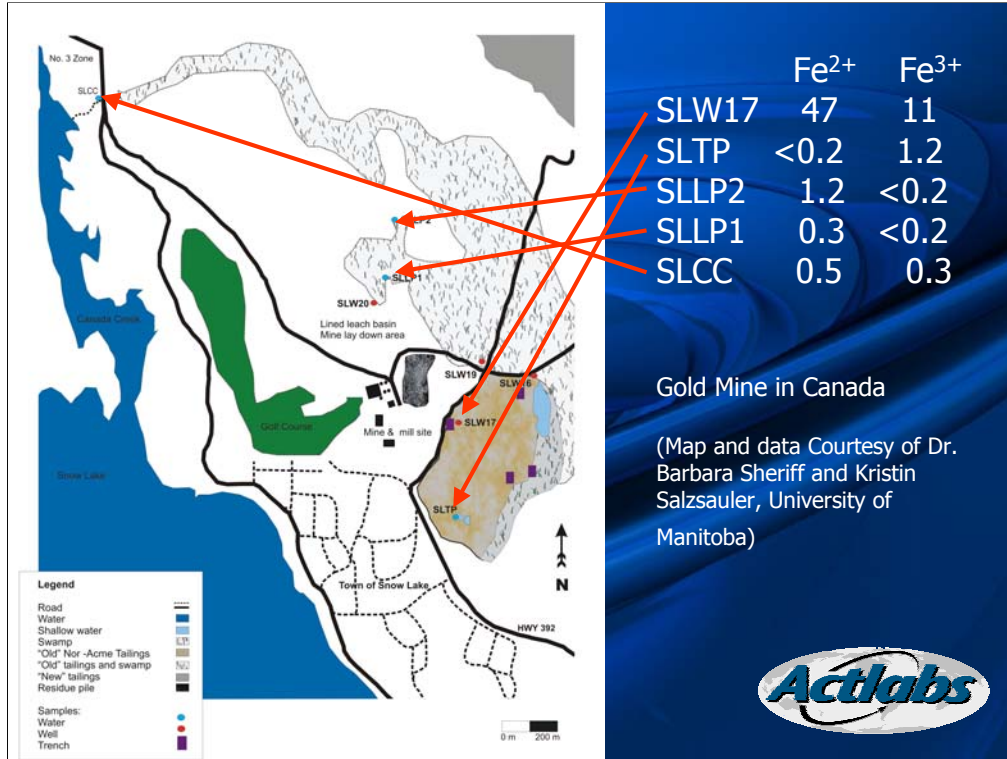
The amount of pore fluid varied from 2-15mls however a very significant number of tests including Ph, Eh, alkalinity, trace metals by ICP/OES and ICP/MS as well as speciation for As and Fe required that our methodology use only a minimum of sample.



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We have also applied this technology to the analysis of stream waters for routine environmental monitoring and are applying the technology up the food chain to biological species.

This technology also promises to have application in mineral exploration where we might expect various species to be present in the vicinity of orebodies that are not generally available in unmineralized areas.



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Conclusions:

- HR-ICP/MS offers one to two orders of magnitude better detection limits than quadrupole ICP-MS
- Resolves most but not all interferences
- Expanded element capability (ie: F)
- Direct vegetation analysis
- PGE and Au in water
- REE in rocks at sub-ppb levels
- Improves precision for Pb isotopic analysis
- Metal speciation for a variety of metals at natural levels is possible by CE-HR-ICP/MS



The first category is simple redox states: the species is a well-defined cation or anion. The second class is organometallic compounds. There is another group for some metals which includes metal complexes: the metal of concern form a coordination bond binding to a ligand.

Monitoring elemental species requires analytical technology with sufficient selectivity and

sensitivity to resolve and quantitate the individual species at ultra trace levels, since an individual toxic species may constitute only a small fraction of an element's total concentration in a sample.

Modern HIGH PRESSURE LIQUID CHROMATOGRAPHY (HPLC) techniques are fairly versatile owing to the diversity of separation technologies available (multiple columns).

The separation mechanism complementary to HPLC is the capillary electrophoresis technique.