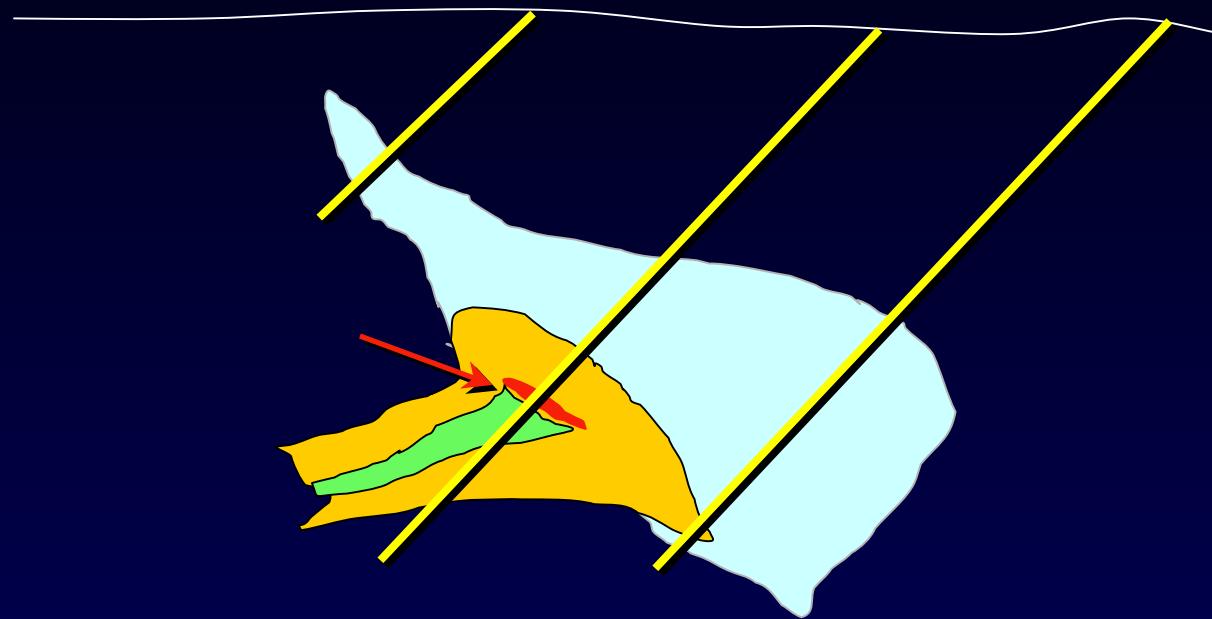


Lithogeochemical Halos: VHMS and SEDEX



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Centre for Ore Deposit Research (CODES)
University of Tasmania

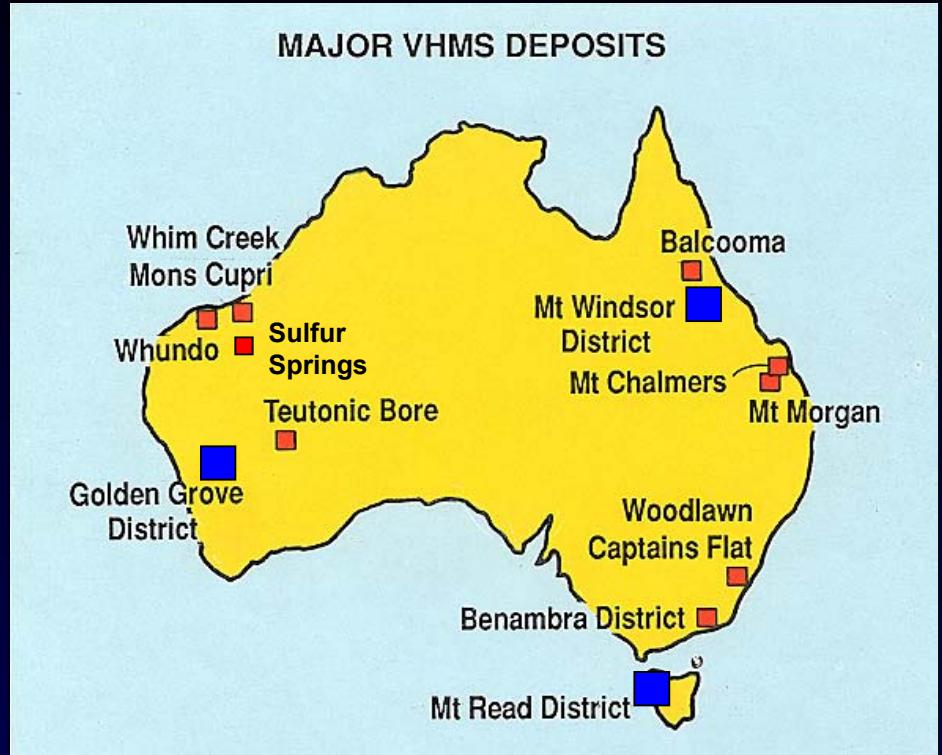


Use of alteration Lithogeochemistry in Exploration

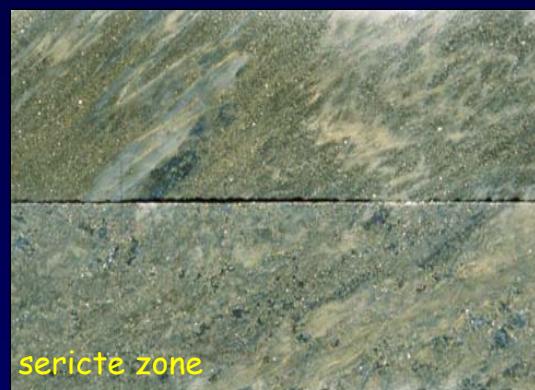
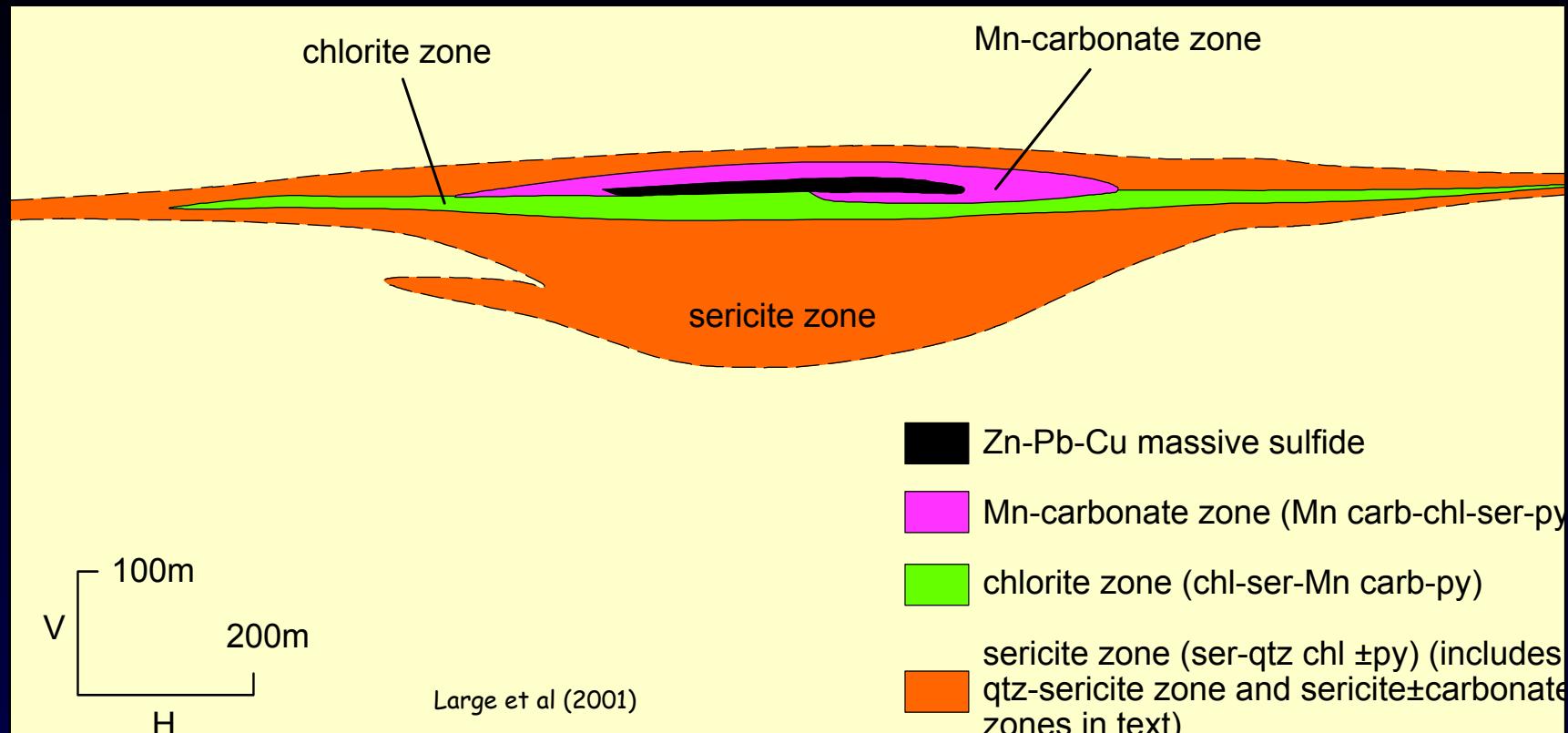
- VHMS deposits (AMIRA P439)
- SEDEX deposits (AMIRA P384 & 384A)

CODES-AMIRA VHMS Alteration Project

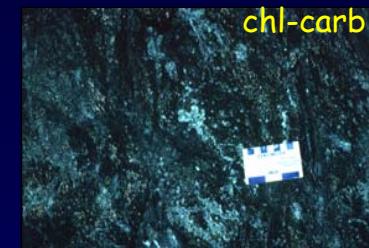
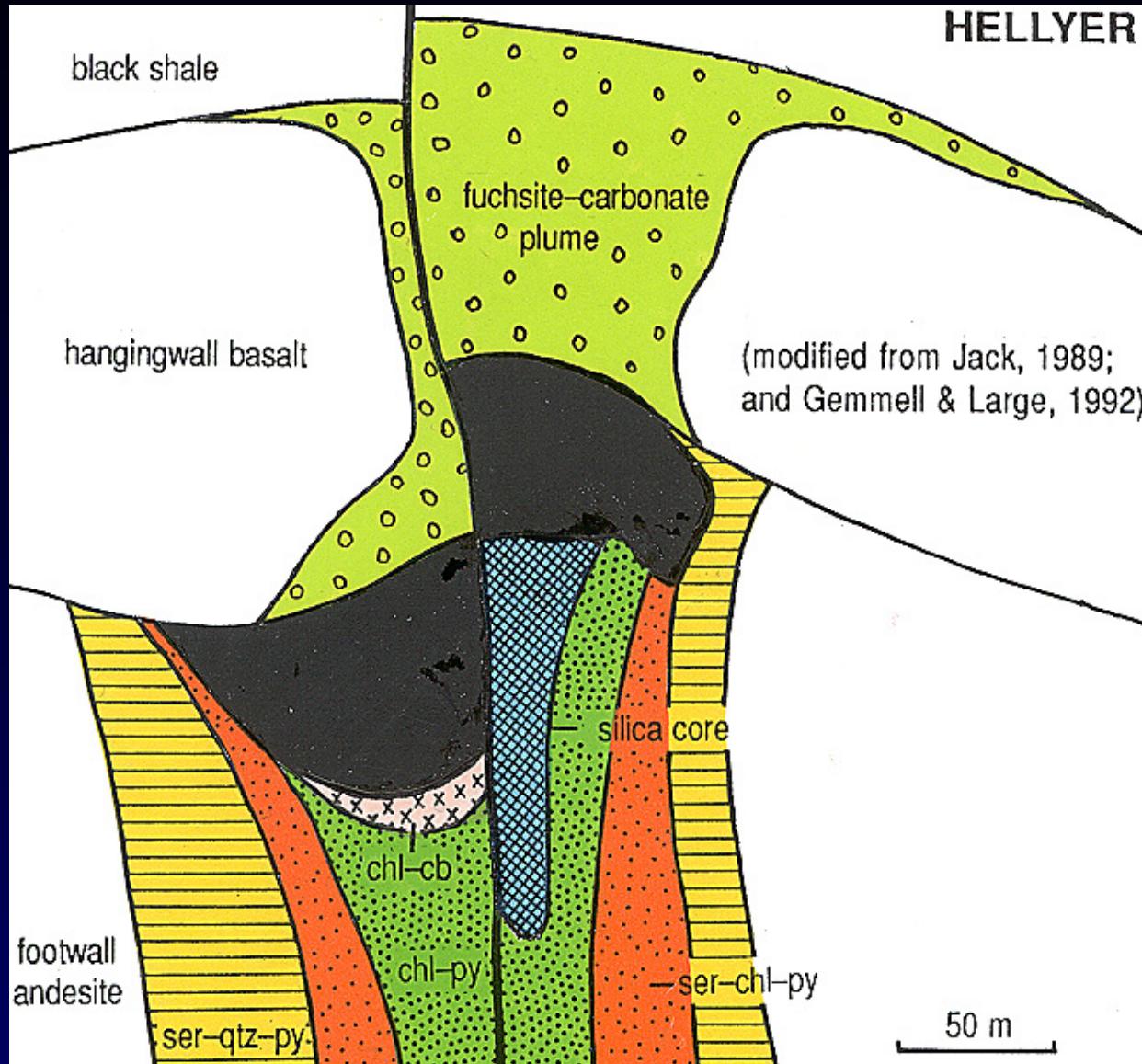
- 3 years
- 11 industry & government sponsors
- 8 staff & 9 PhD/MSc students
- Regional and deposit scales



Rosebery Cu-Zn-Pb: Alteration Zonation



Hellyer Zn-Pb-Cu -Alteration Zonation



The Alteration Box Plot for VHMS Systems

- Based on the Ishikawa alteration index developed for Kuroko deposits
- $$AI = \frac{100(MgO + K_2O)}{(MgO + K_2O + Na_2O + CaO)}$$

elements added by alteration
chlorite & sericite alteration

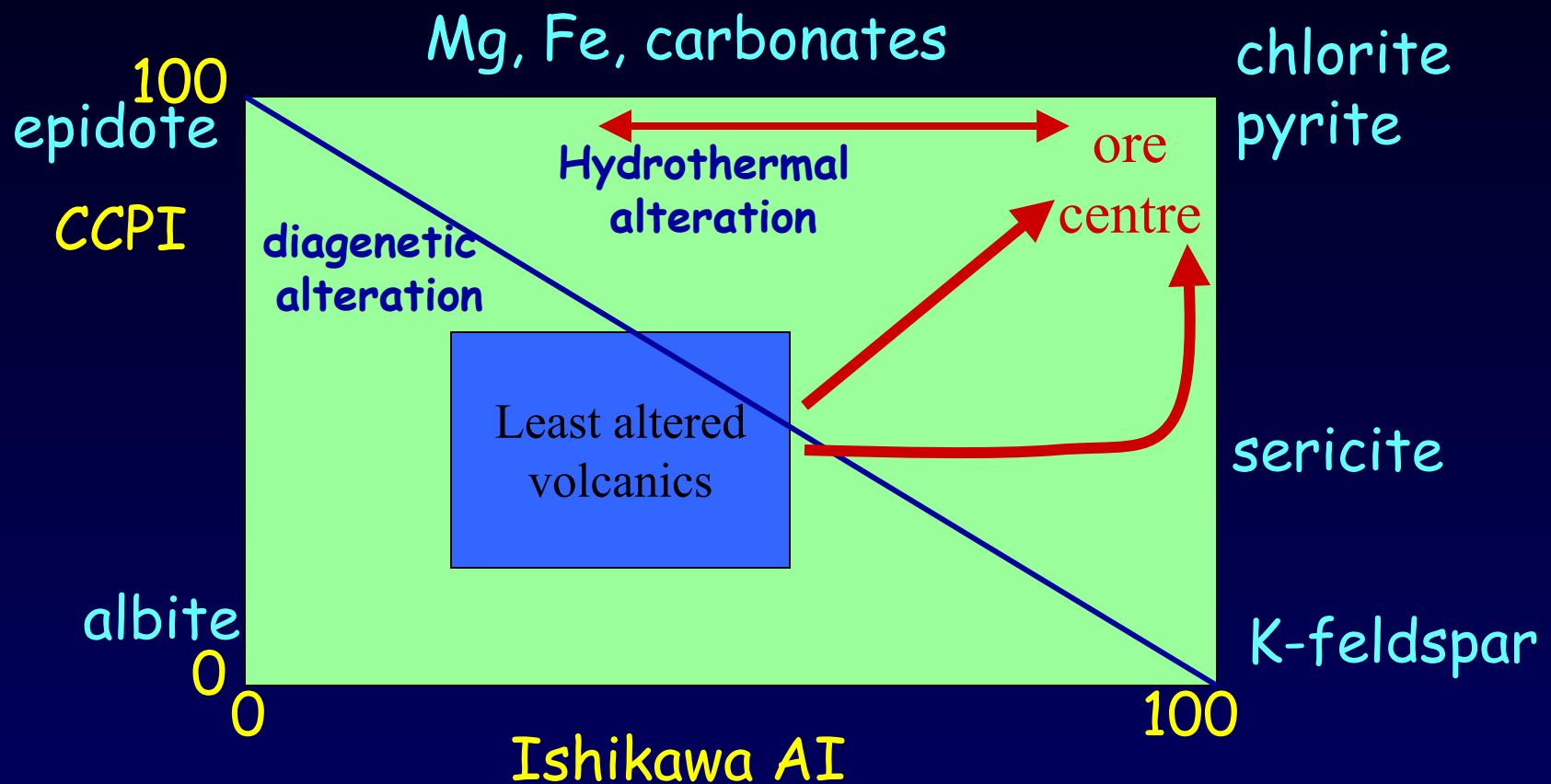
elements subtracted by alteration
replacement of plagioclase & glass
- Varies from ~40 (unaltered) to 100 (altered)

Other VHMS vector

- Chlorite-carbonate-pyrite index: CCPI
- Measures the intensity of these proximal alteration minerals
- $$CCPI = \frac{100(MgO + FeO^*)}{(MgO + FeO^* + Na_2O + K_2O)}$$
- Enables the separation of chlorite, sericite and carbonate alteration.

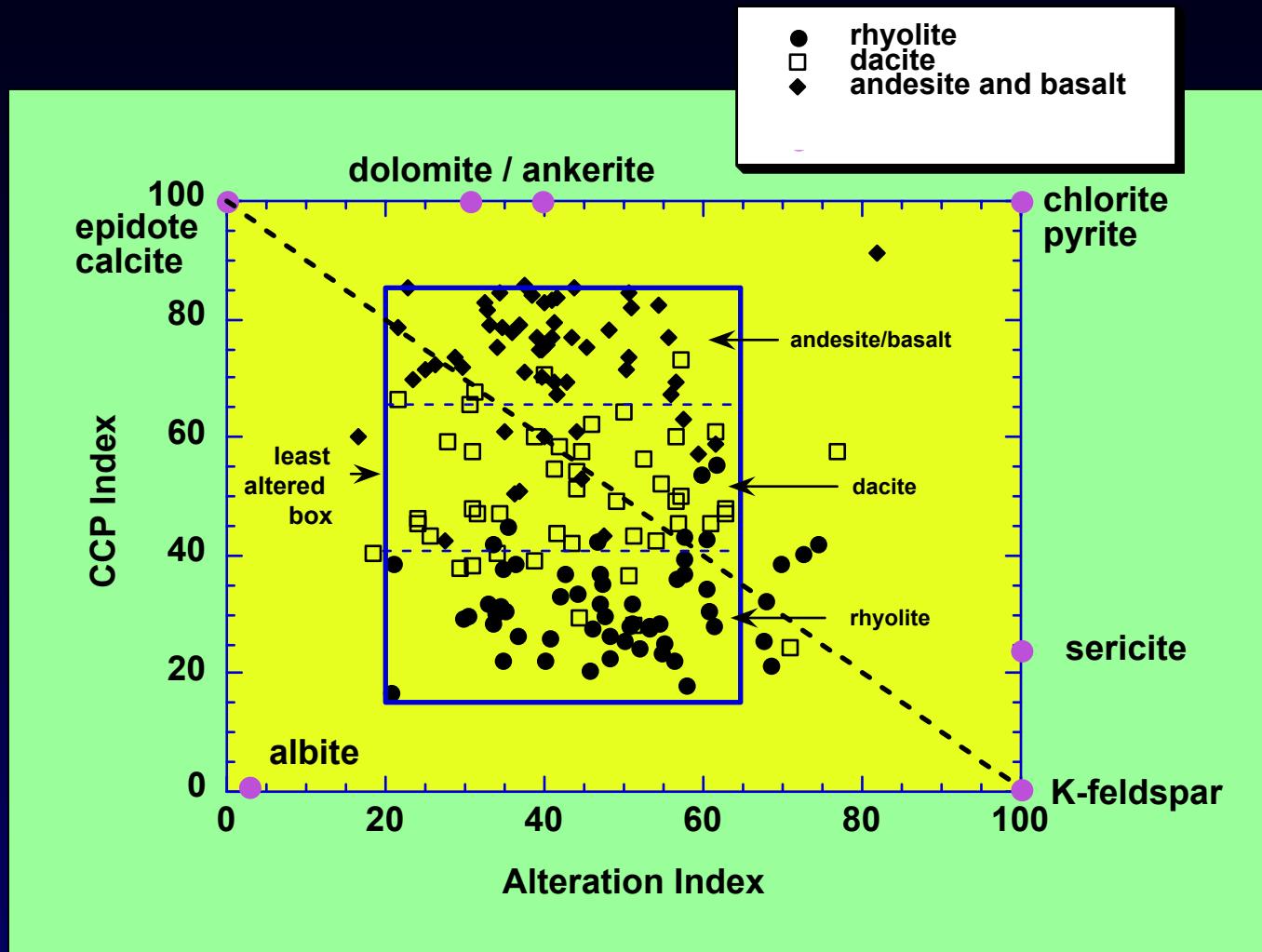
VHMS Alteration Box Plot

Large, Gemmell, Herrman, Paulick & Huston (2001)

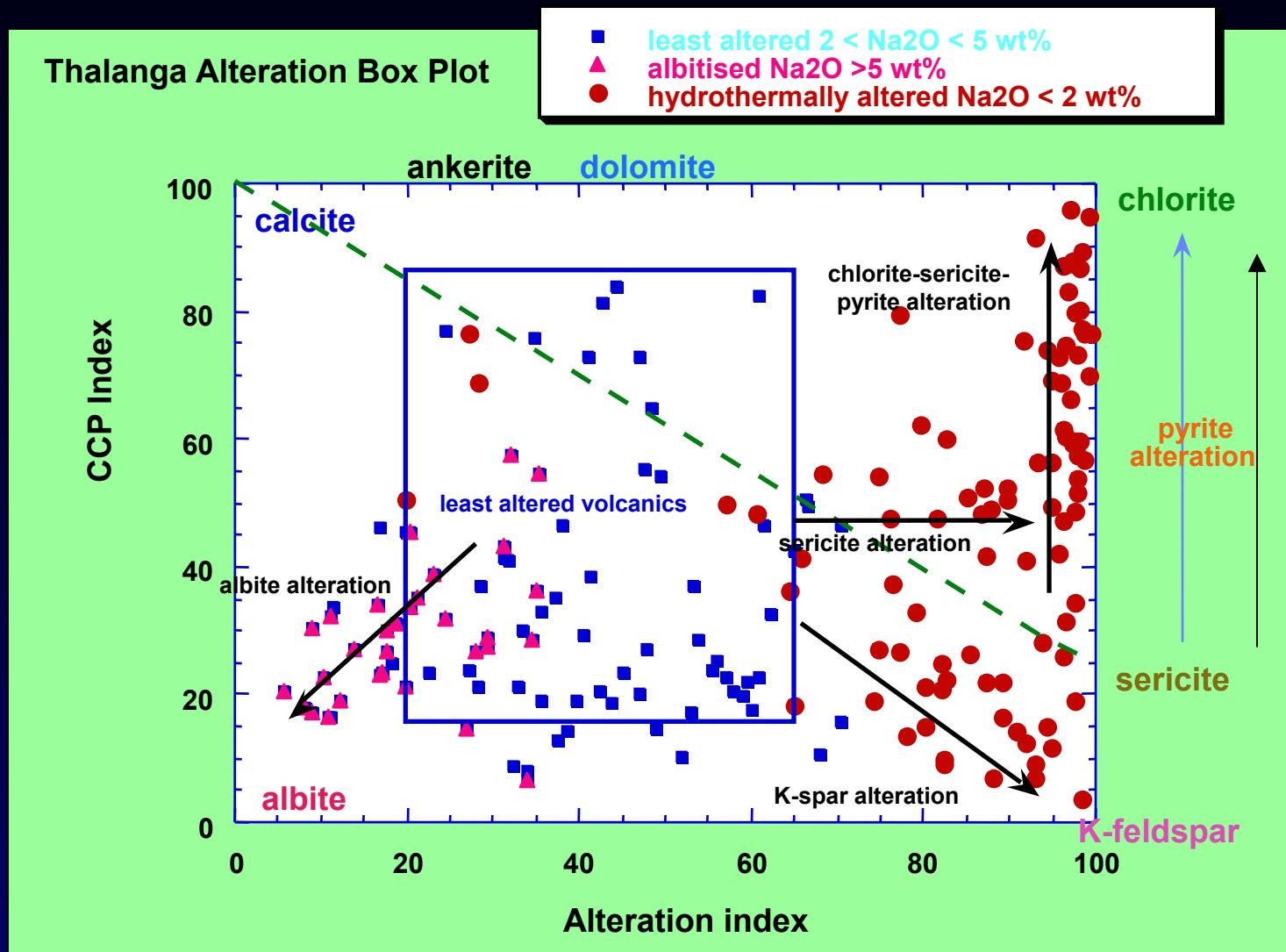


Least altered volcanics

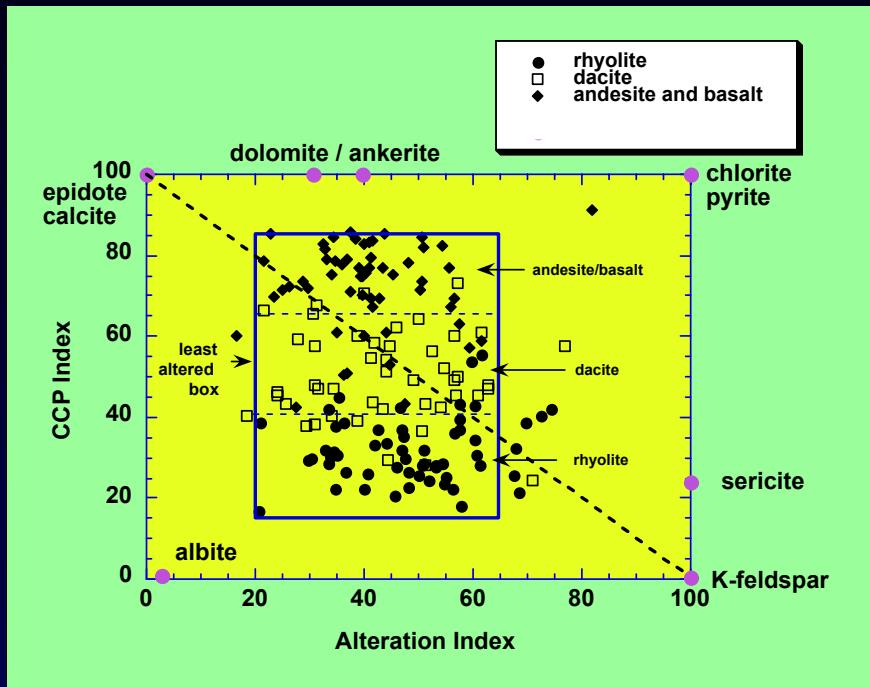
Mt Read Volcanic Belt



Example: Thalanga VHMS



Advantages of alteration vector plots



- Simple to apply
- Defines least altered rocks
- Relates geochemistry to mineralogy
- Shows alteration trends
- Defines very weak alteration
- Distinguishes hydrothermal alteration from diagenetic and metamorphic “alteration”

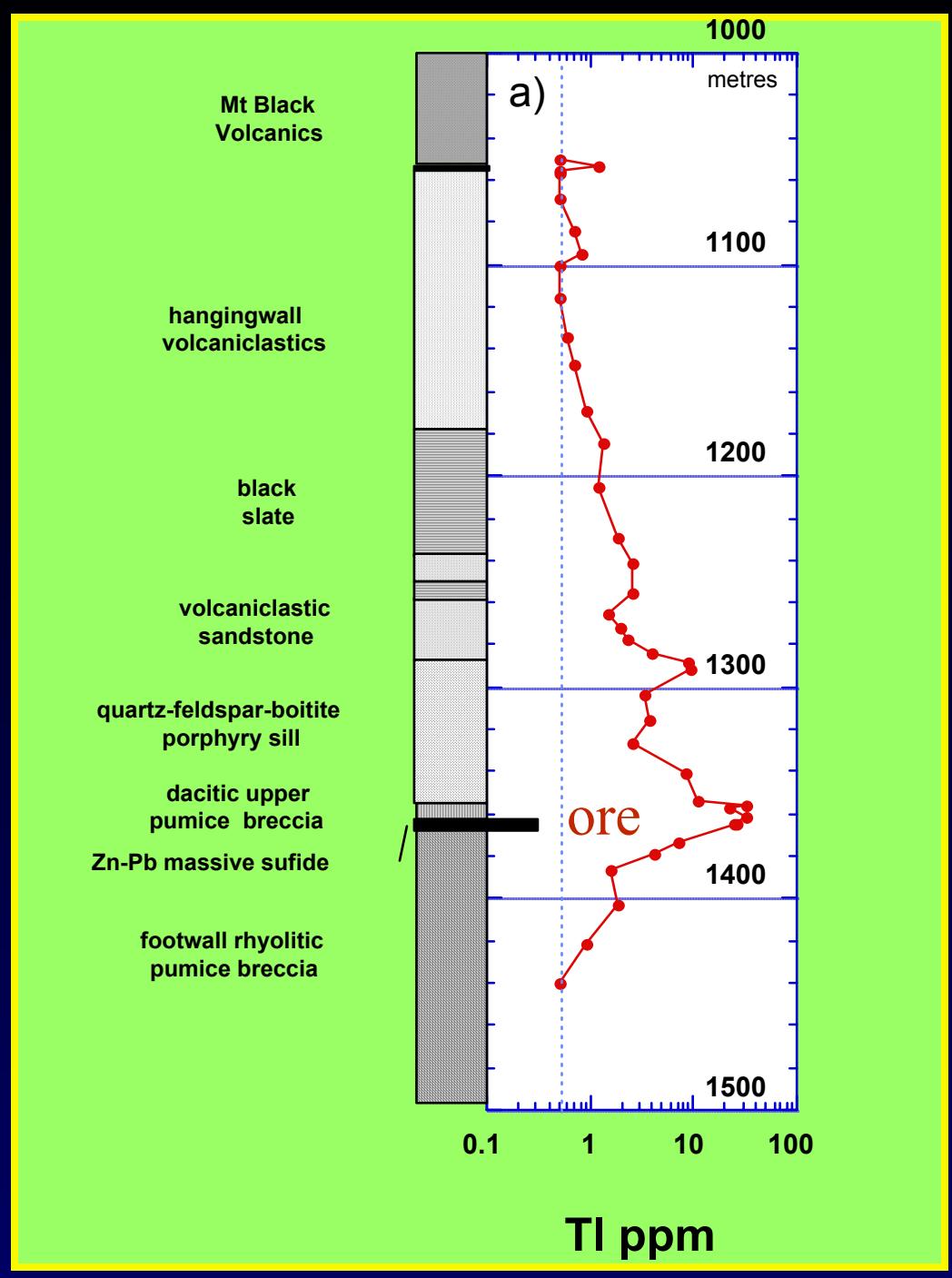
Best Halos and Vectors

- Ishikawa AI
- Mn
- S/Na₂O
- Ba/Sr
- Tl and Sb

Increasing
size

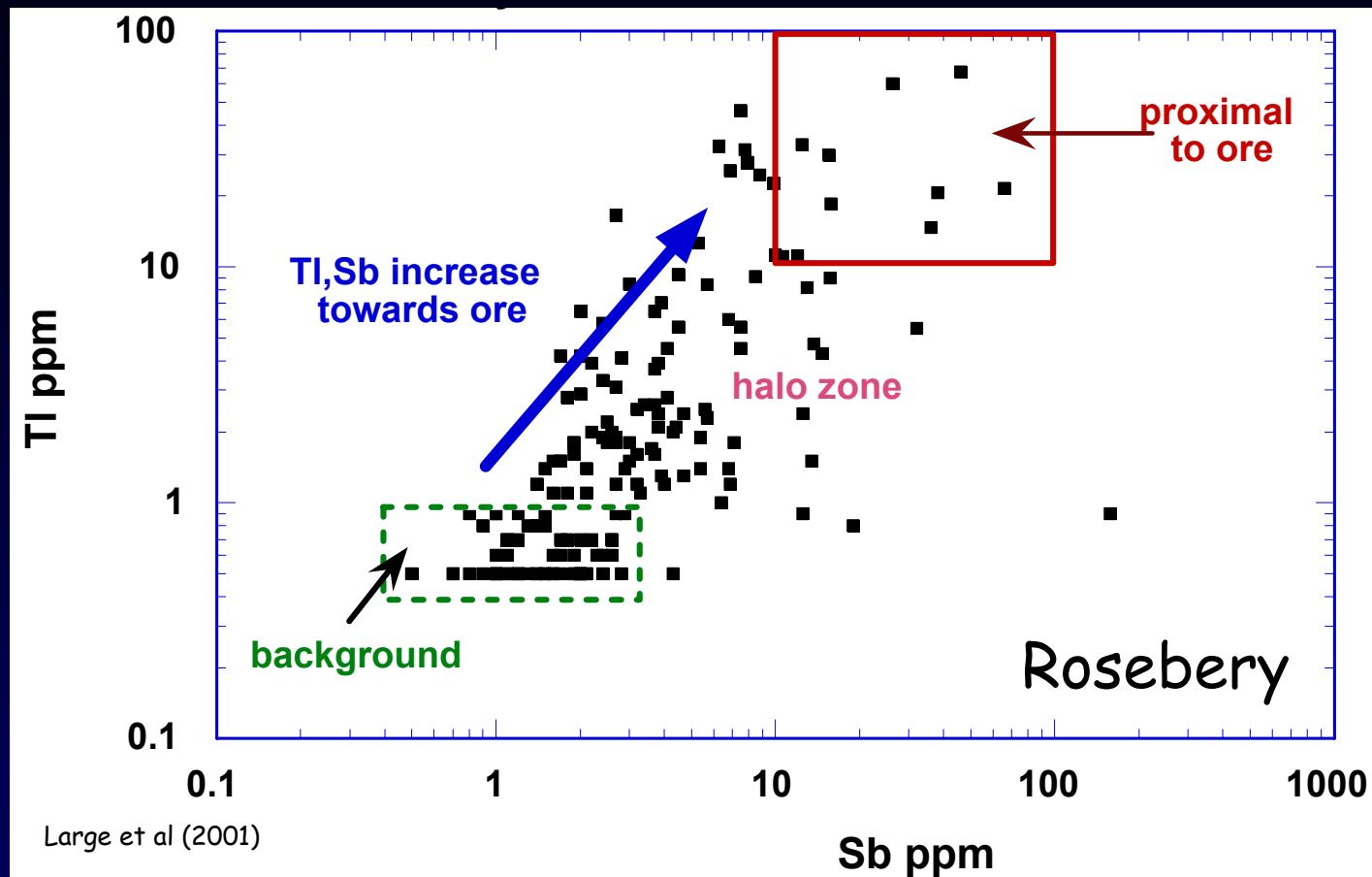


Rosebery thallium halo



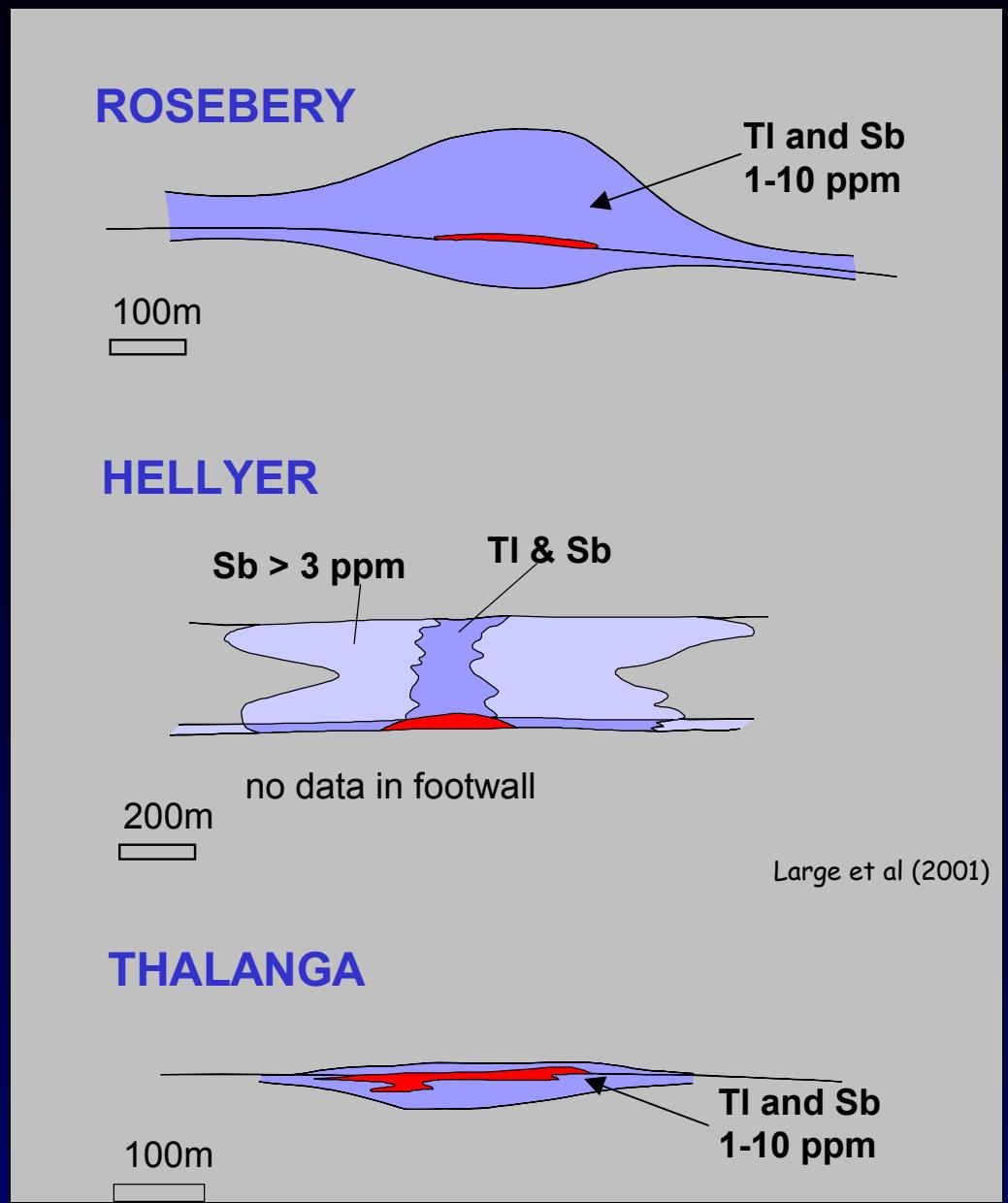
Thallium and Antimony Halos

- very useful vectors for polymetallic VHMS deposits

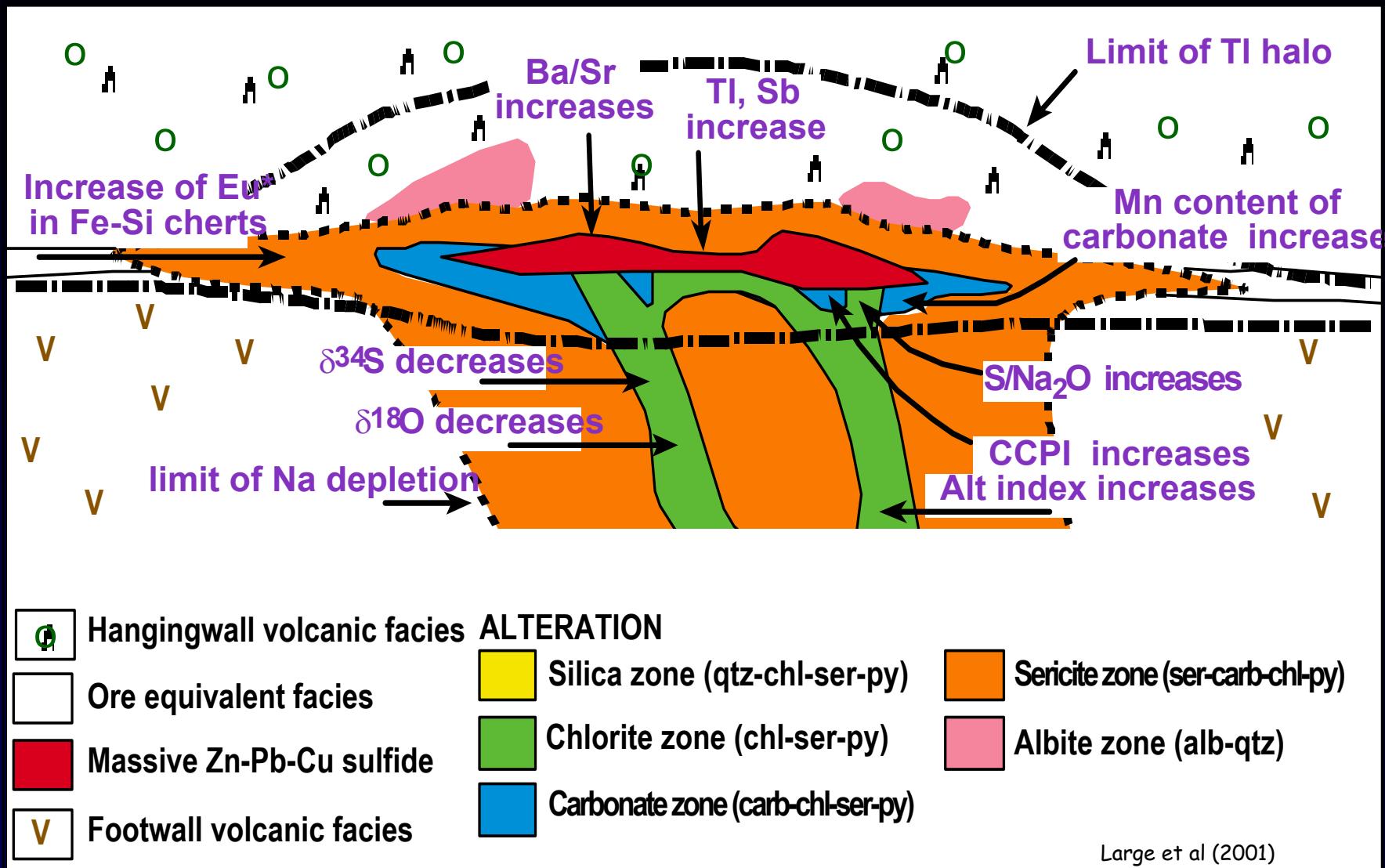


Thallium and Antimony Halos

- Up to 100 ppm Tl and Sb proximal to ore and 1-10 ppm within halo zone
- Sb and Tl halo within favourable horizon and extending vertically and laterally into hangingwall
- Tl (> 1 ppm) up to 50 m into HW and FW and along favourable horizon



Zn-rich Polymetallic VHMS Deposits

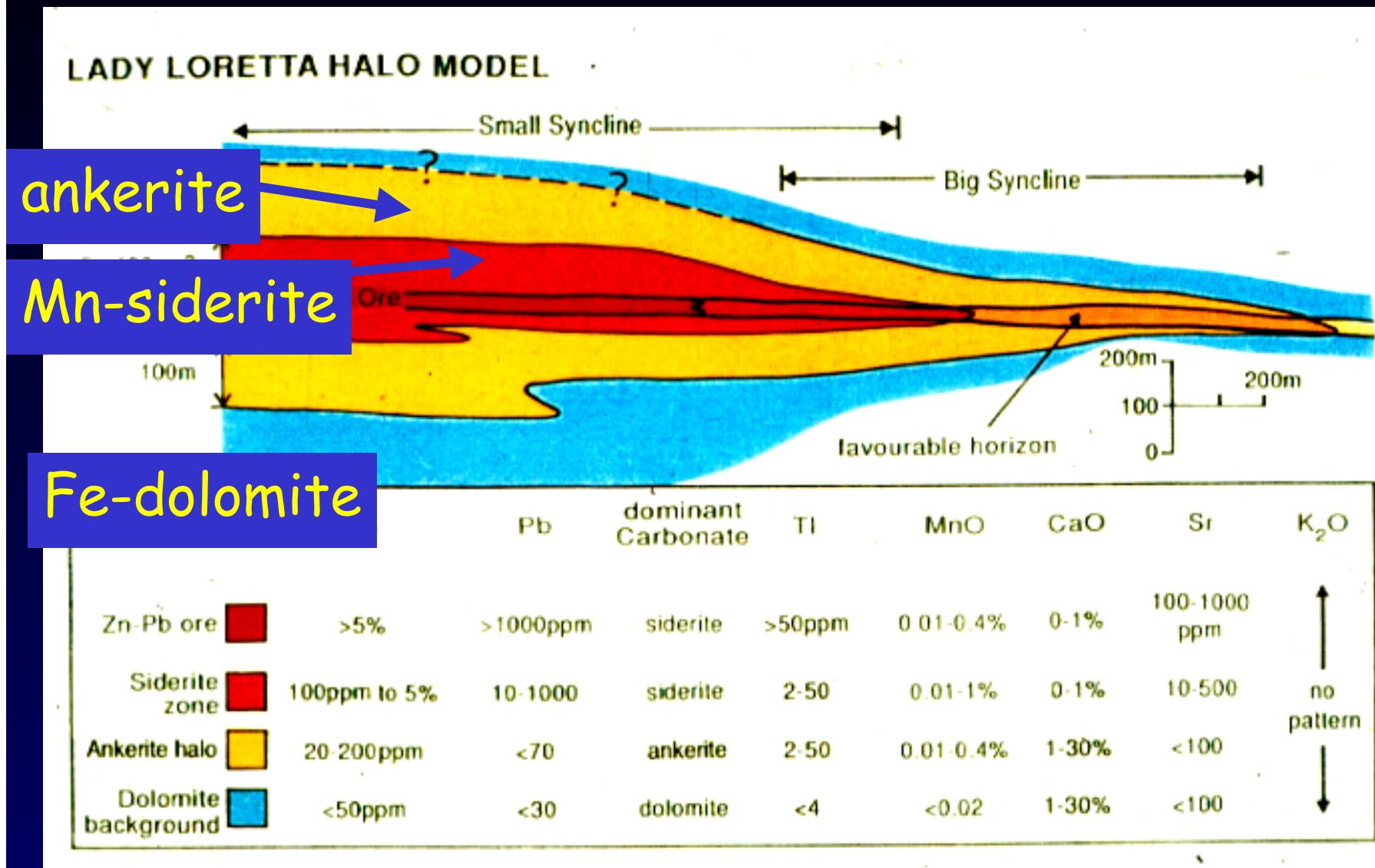


Halos and Vector Diagrams for SEDEX Systems (AMIRA P384)

- Based on research at Lady Loretta and HYC
- Controlled by the change in chemistry of carbonate minerals as you approach the orebody.
- Dolomite -> ferroan dolomite -> ankerite -> Mn-siderite -> Zn-Pb ore



Fe-Mn-carbonate halo model



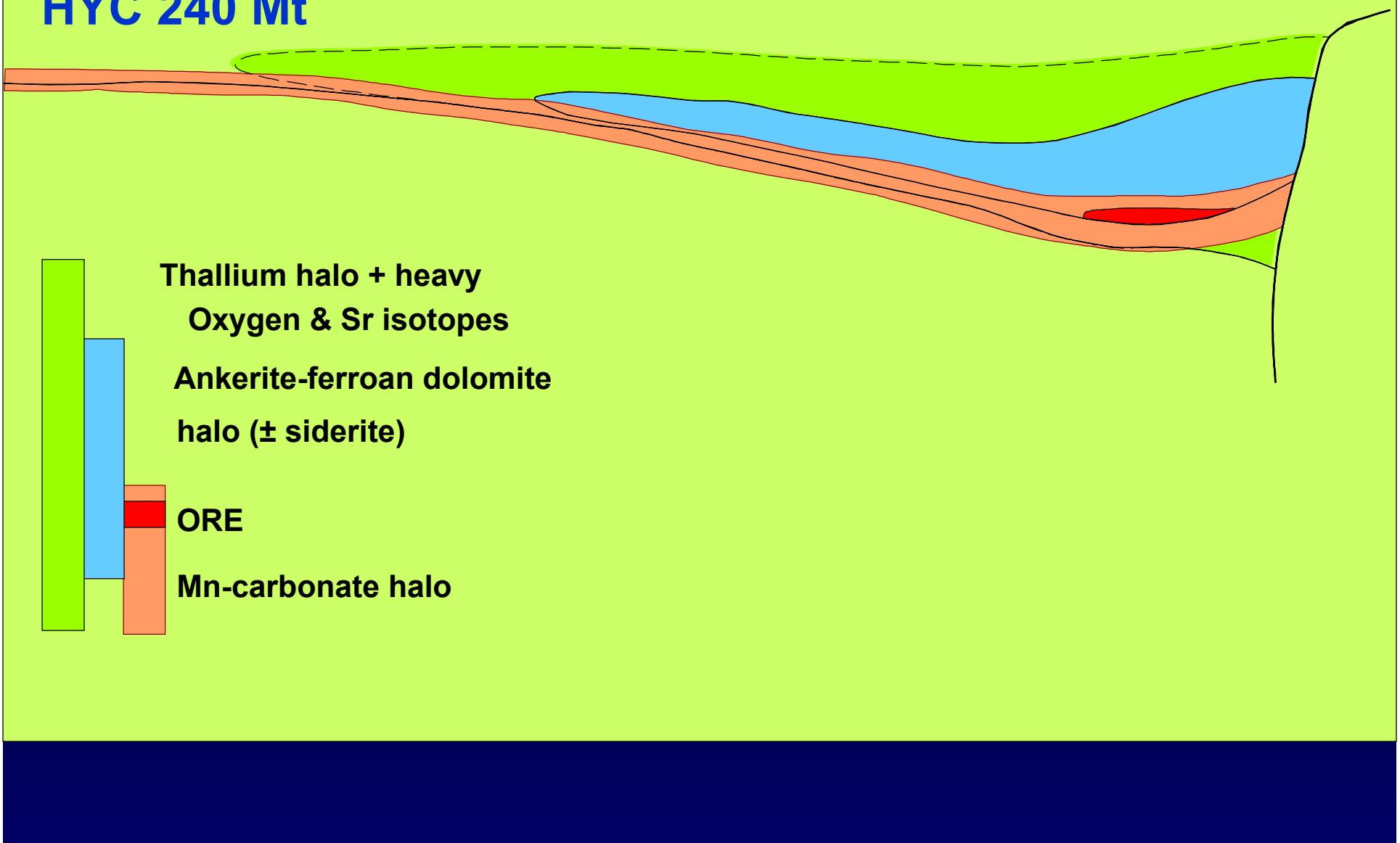
Siderite halo in siltstone host rocks





Sedex Zn-Pb-Ag footprint of overlapping halos

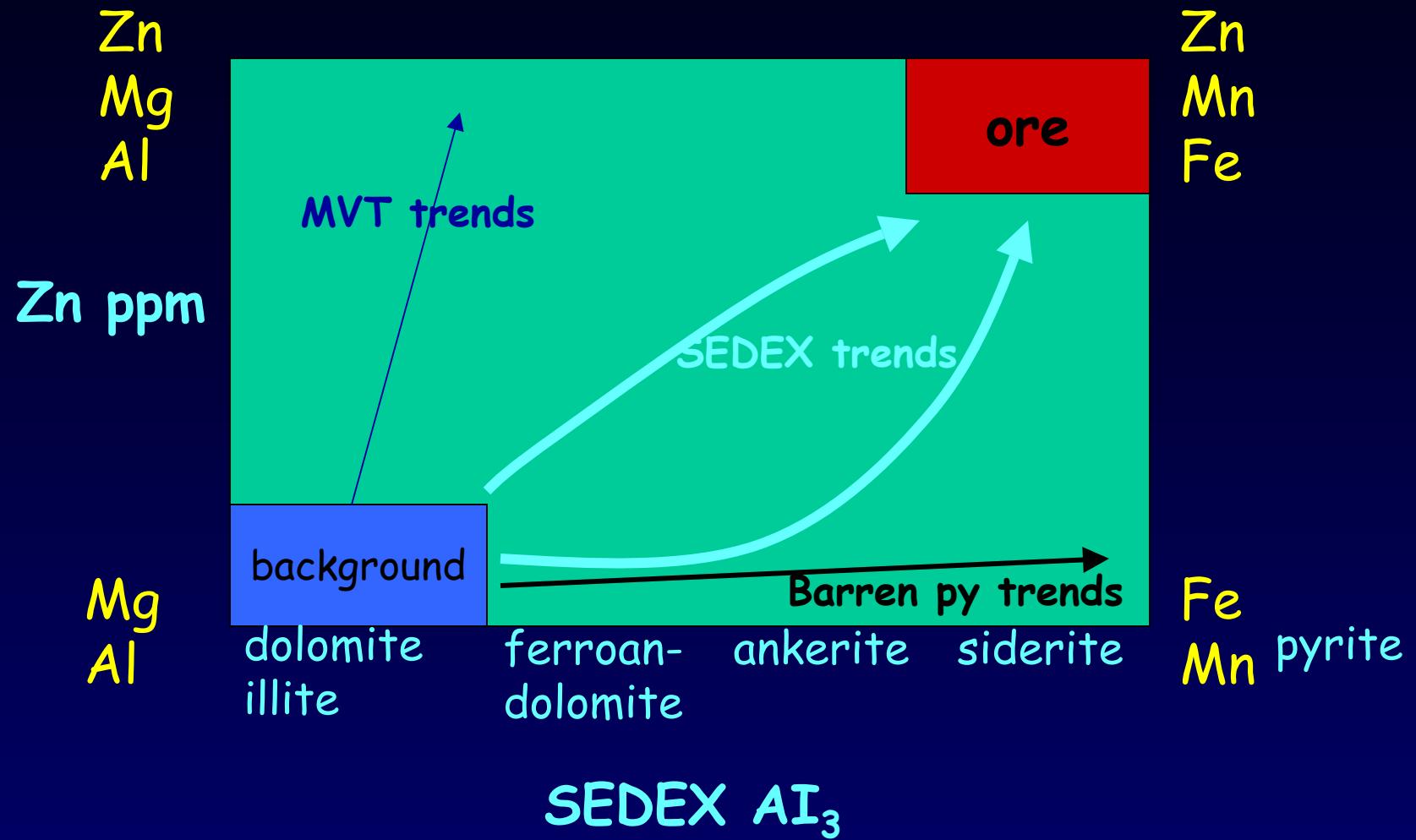
HYC 240 Mt



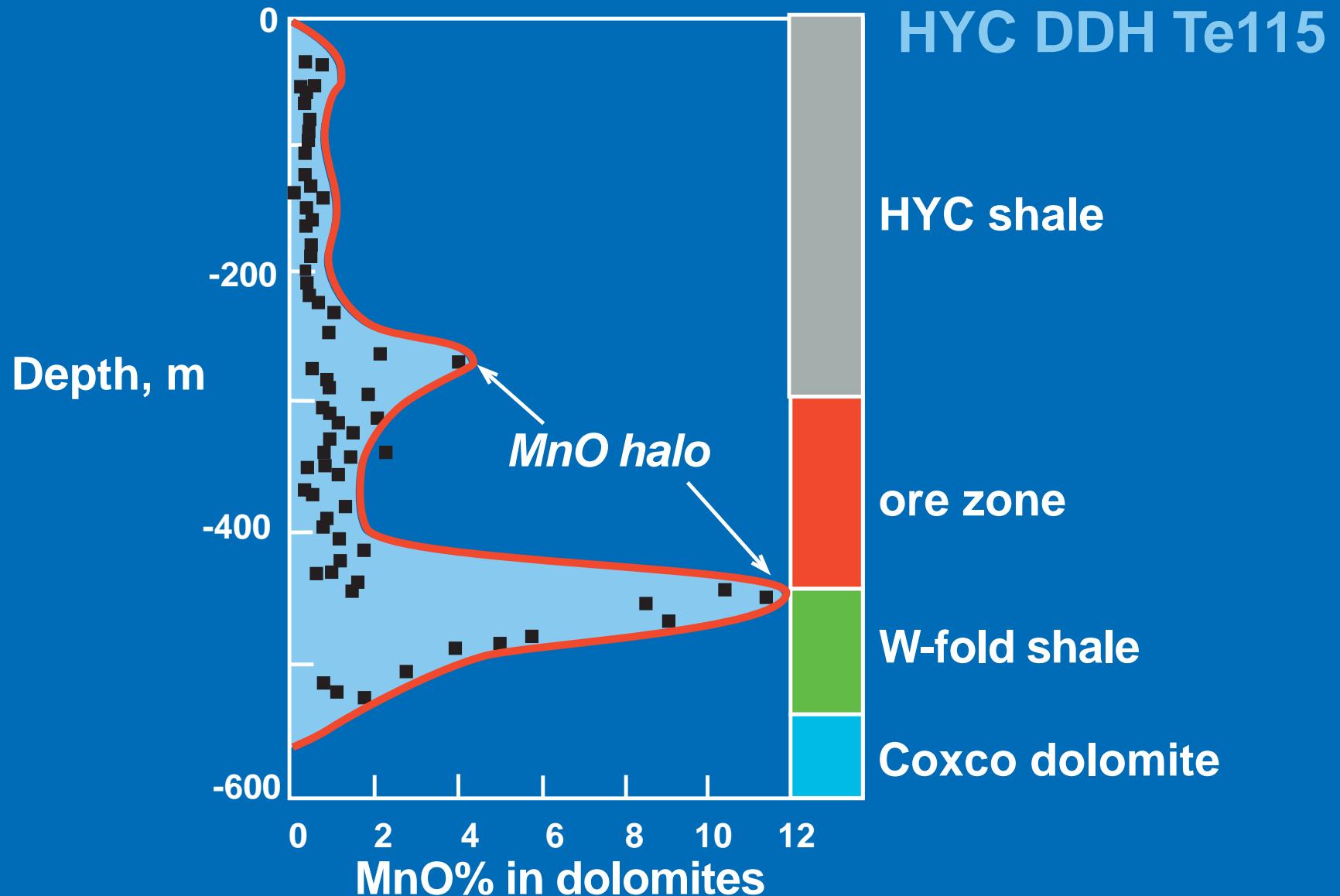
Key Vectors

- SEDEX AI₃ = $\frac{100(\text{FeO}^* + 10\text{MnO})}{(\text{FeO}^* + 10\text{MnO} + \text{MgO} + \text{Al}_2\text{O}_3)}$
- MnO_d = $\frac{\text{MnO}^* 40.03}{\text{CaO}}$ (MnO content in dolomite)
- Ti, C/O isotopes in carbonates, Sr isotopes

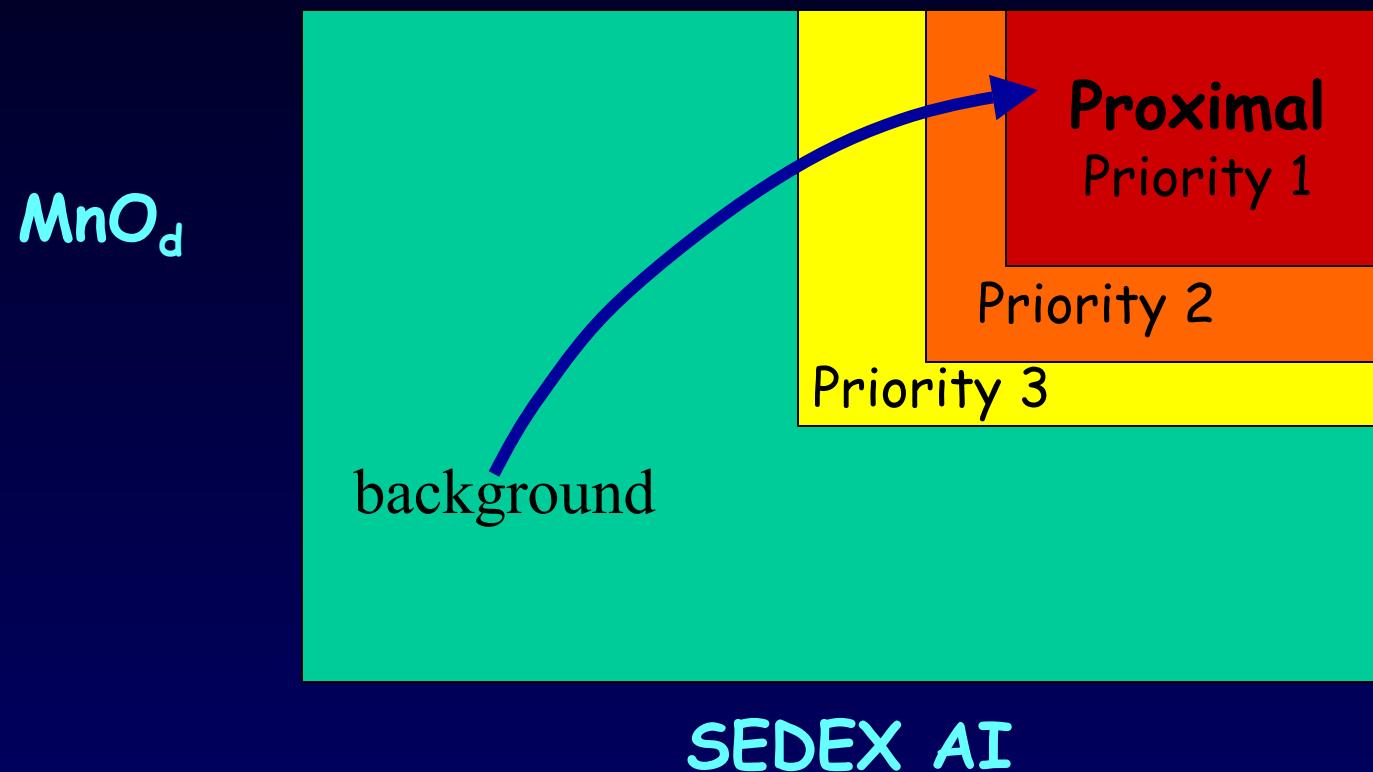
SEDEX Vector Plots



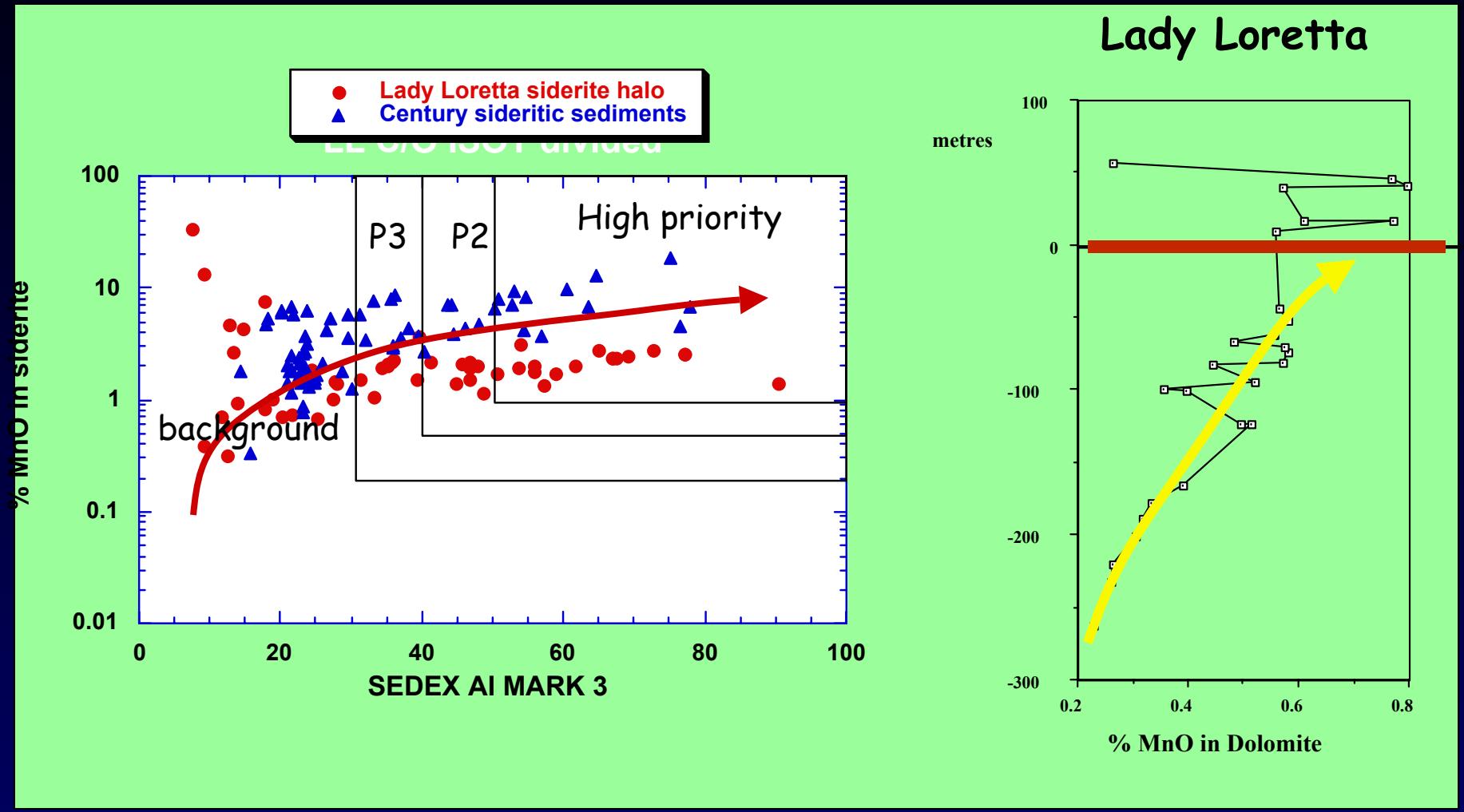
Mn content in carbonate



SEDEX Vector Plots



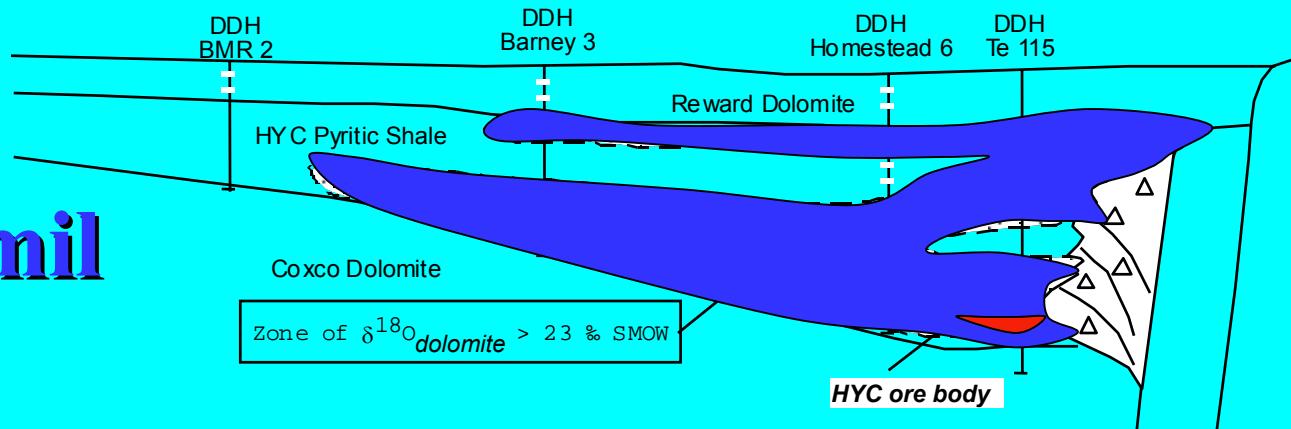
Vector plots for SEDEX



C-O isotope halos at HYC

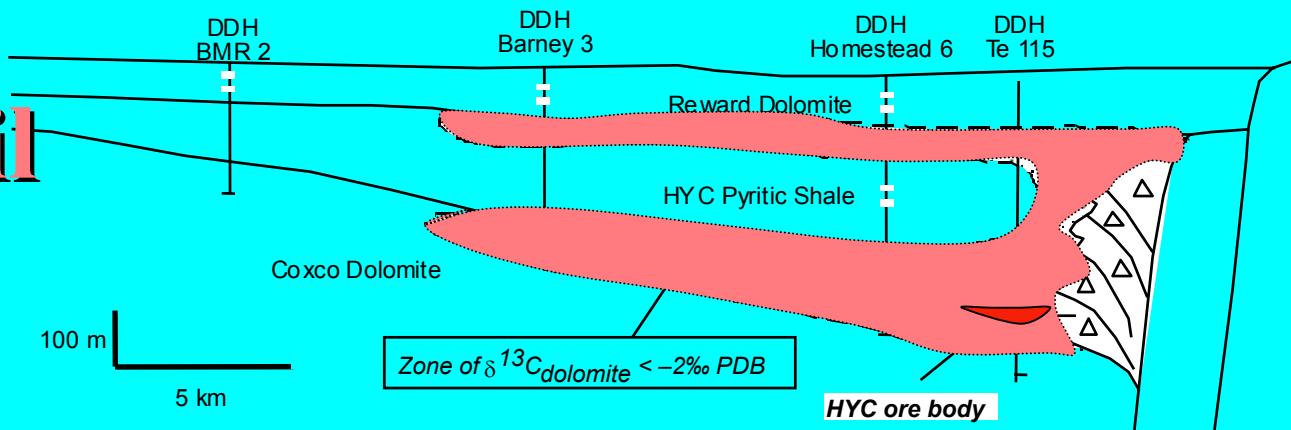
Oxygen
> 22.5 permil

A Oxygen isotope halo

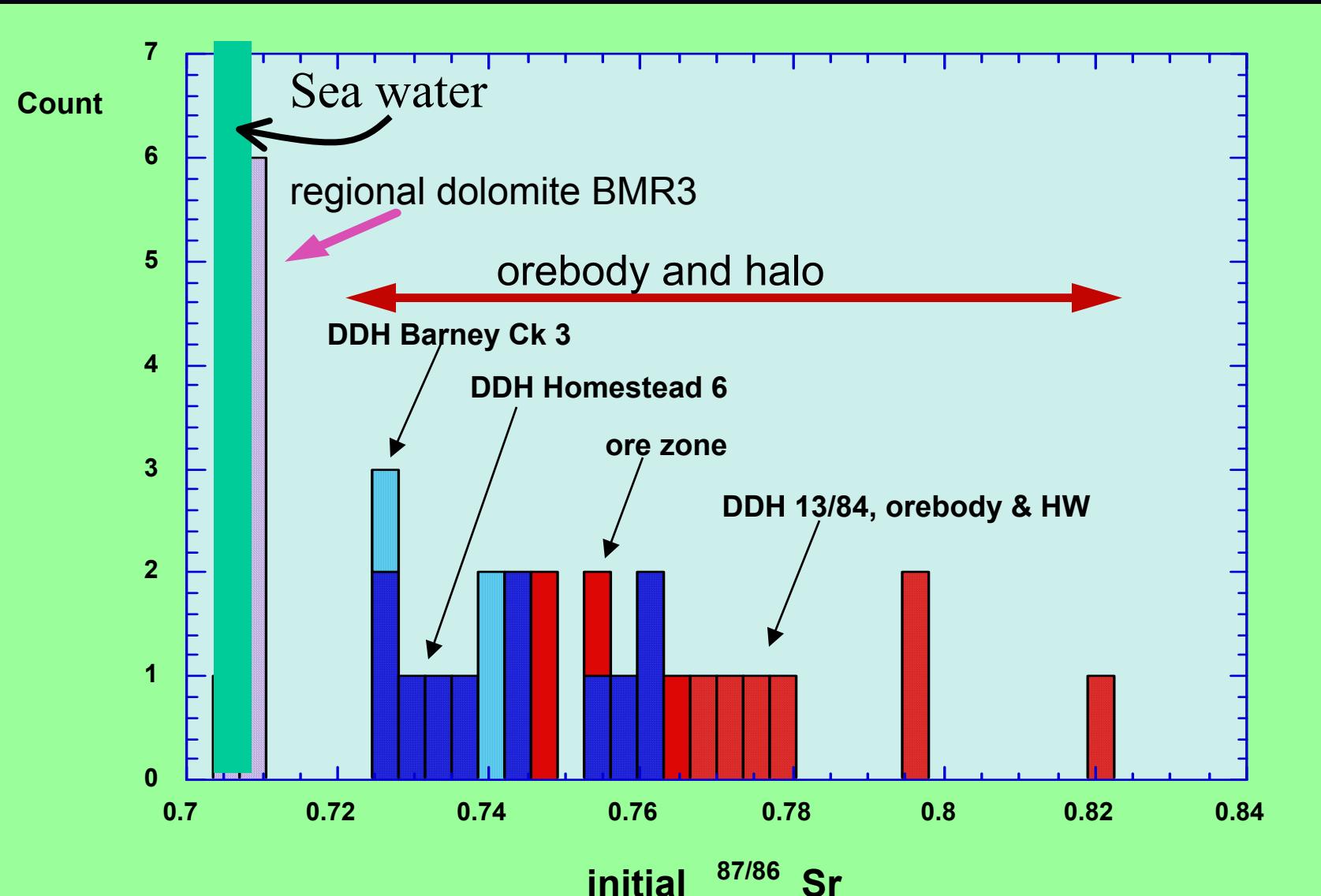


Carbon
< -2 permil

B. Carbon isotope halo



Initial $^{87}/^{86}\text{Sr}$ range



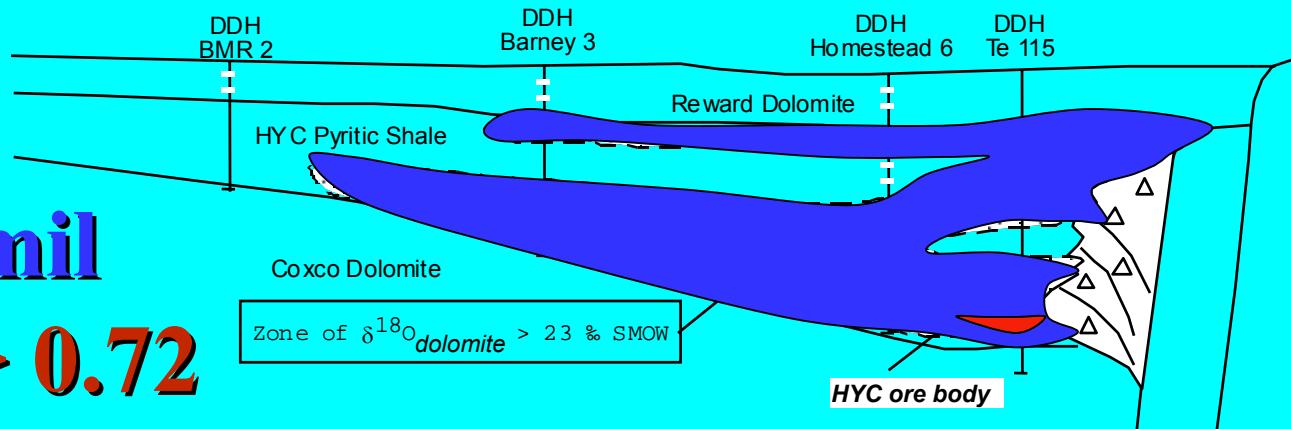
C-O isotope halos at HYC

Oxygen

>22.5 permil

$^{87}\text{Sr}/^{86}\text{Sr} > 0.72$

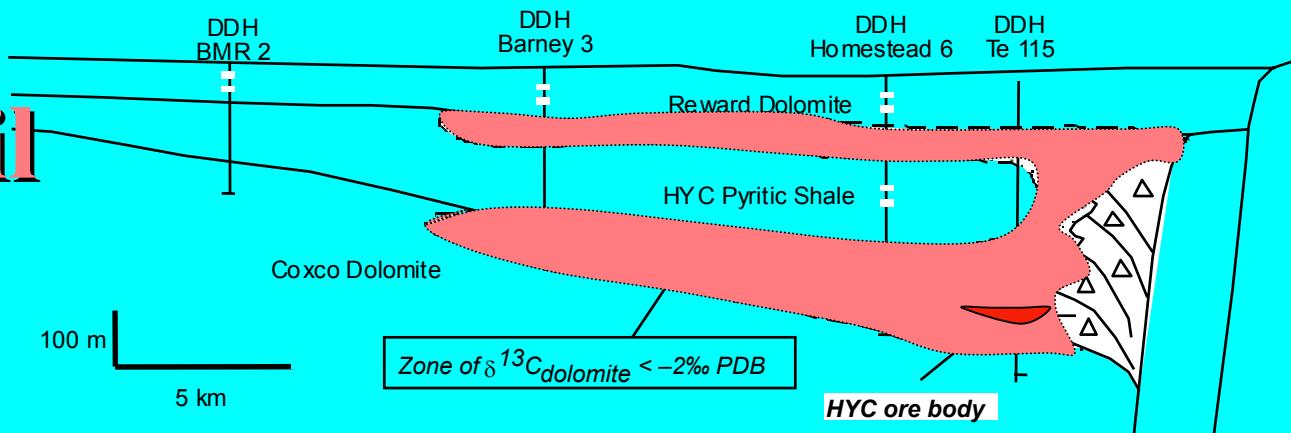
A Oxygen isotope halo



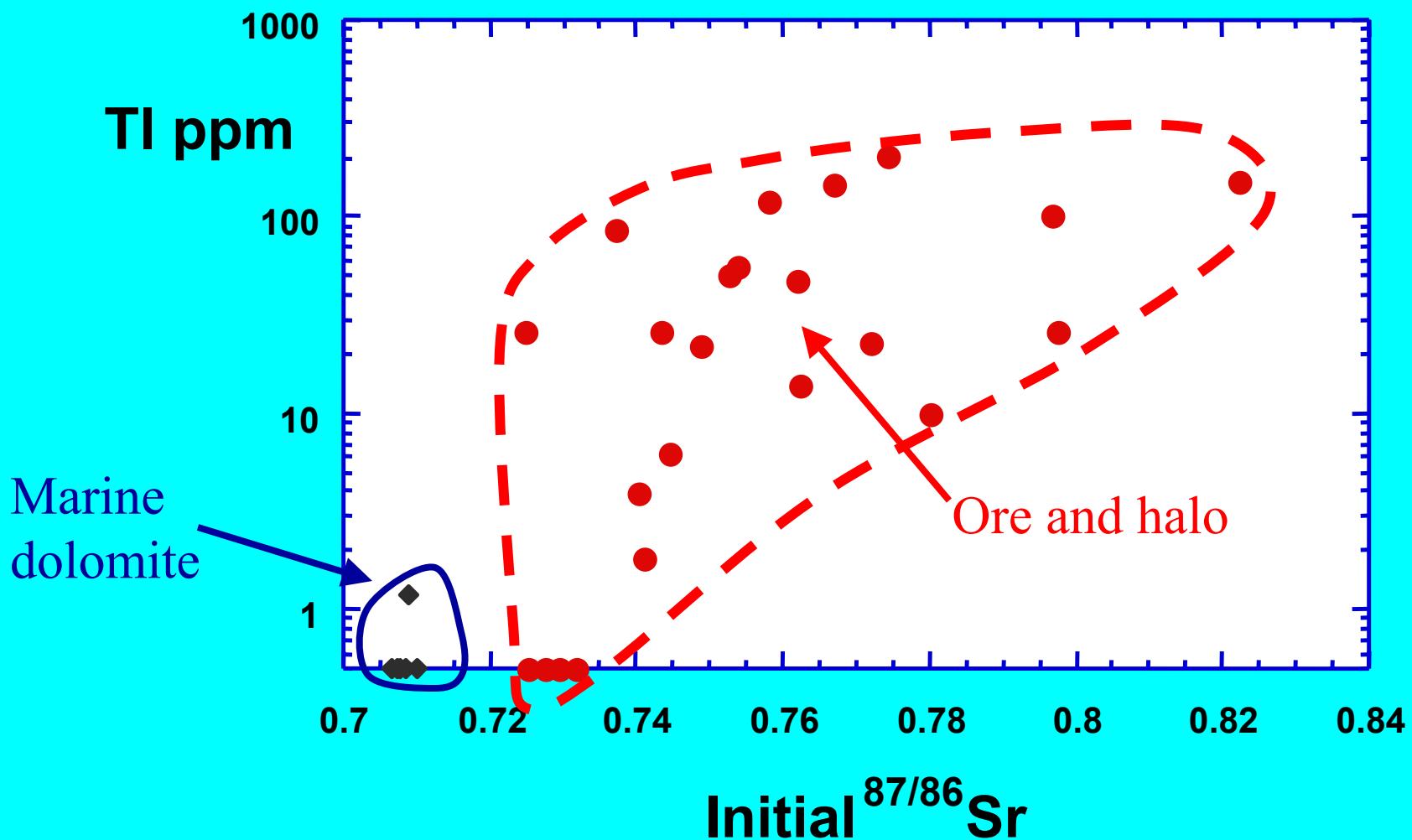
Carbon

< -2 permil

B. Carbon isotope halo

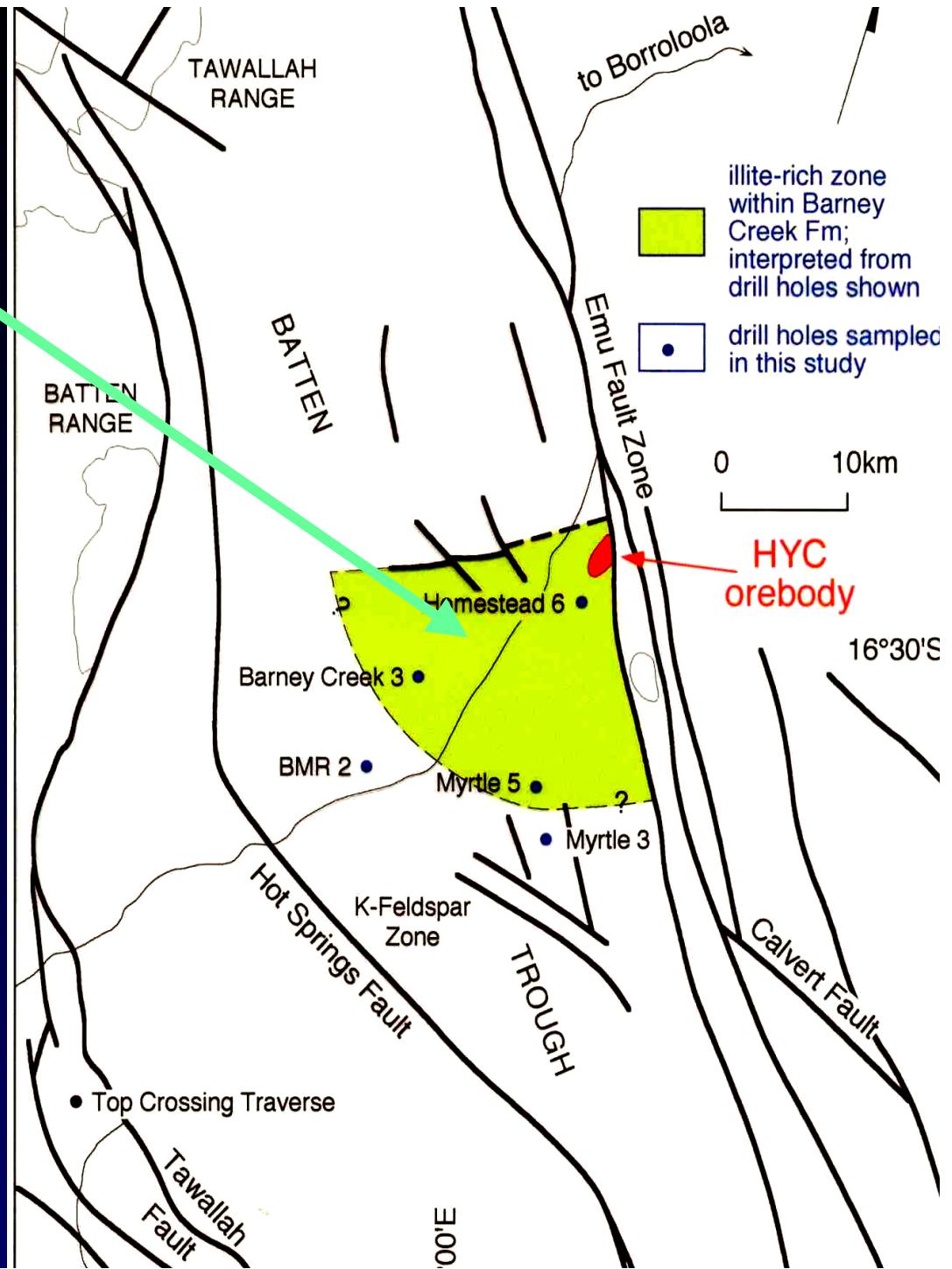


Thallium- Sr isotope halo



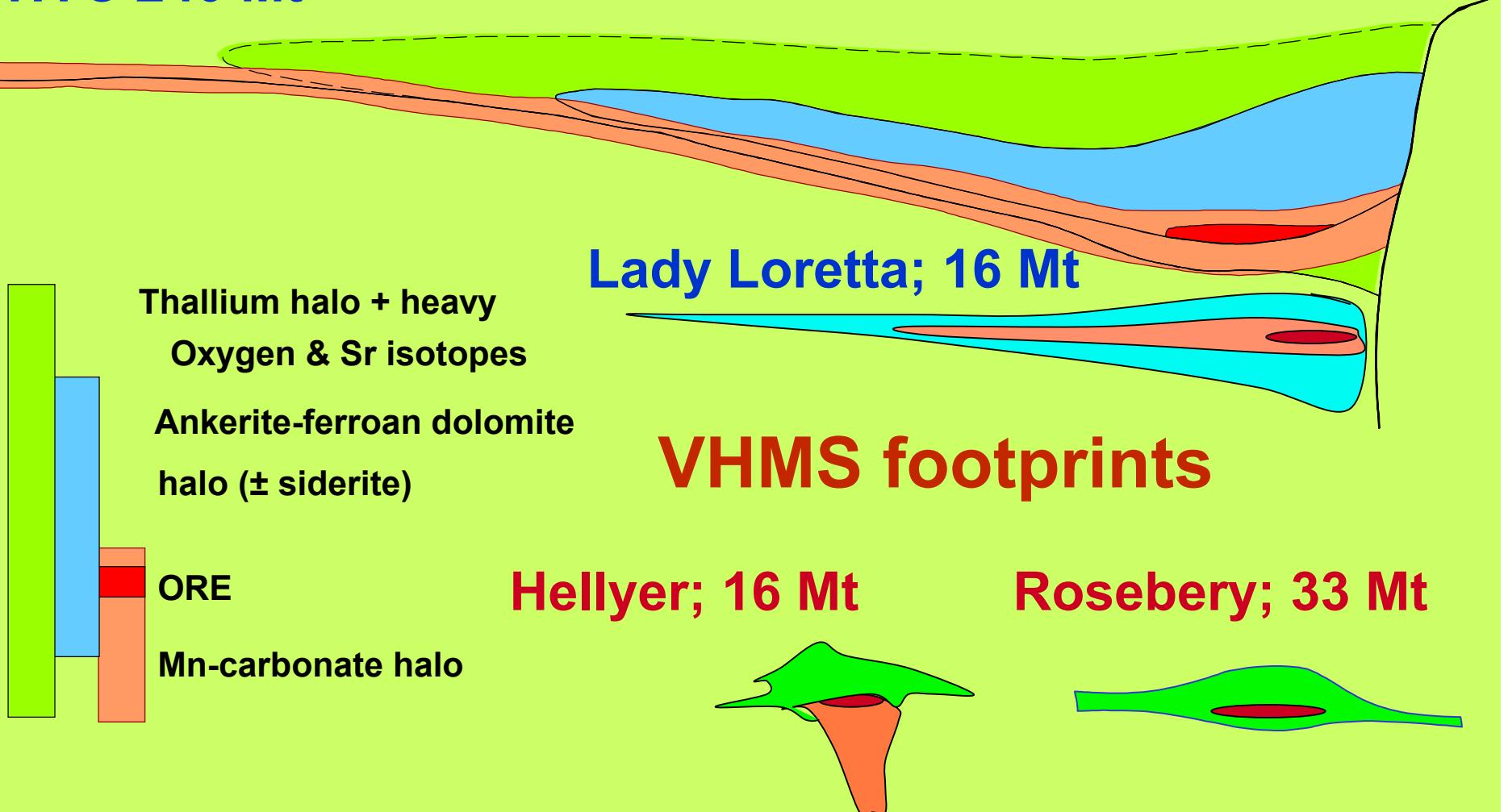
HYC Halo

- Sedex AI > 50
- MnO_d > 1.5 wt%
- Tl > 4 ppm
- δ¹⁸O > 22.5 permil
- δ¹³C < -2 permil
- 87/86Sr > 0.7200



Sedex Zn-Pb-Ag footprints of overlapping halos

HYC 240 Mt



END