Mapping of Mineralized Groundwater Discharge into Lakes and Rivers

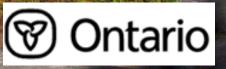
Blair Hostetler James Ray Richard Dyer Gary Grabowski David Guindon Ross Kelly

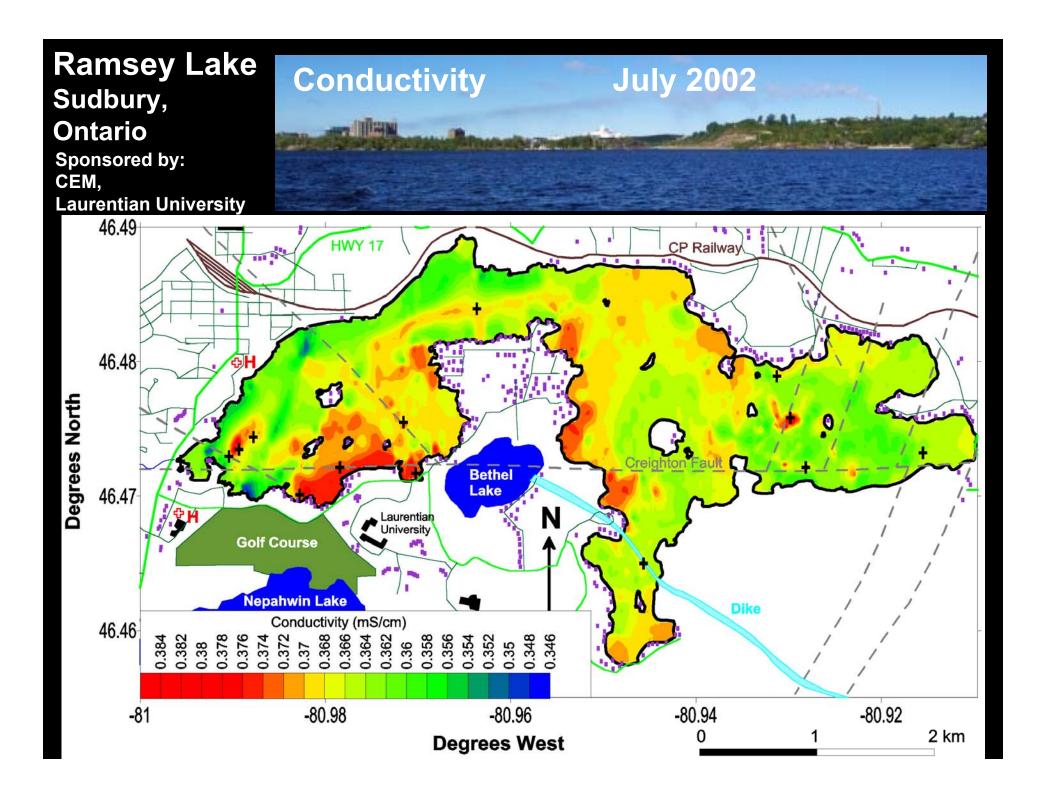
Aquapath Pty Ltd Aquapath Pty Ltd Ontario Geological Survey Ontario Geological Survey Ontario Geological Survey Ontario Geological Survey



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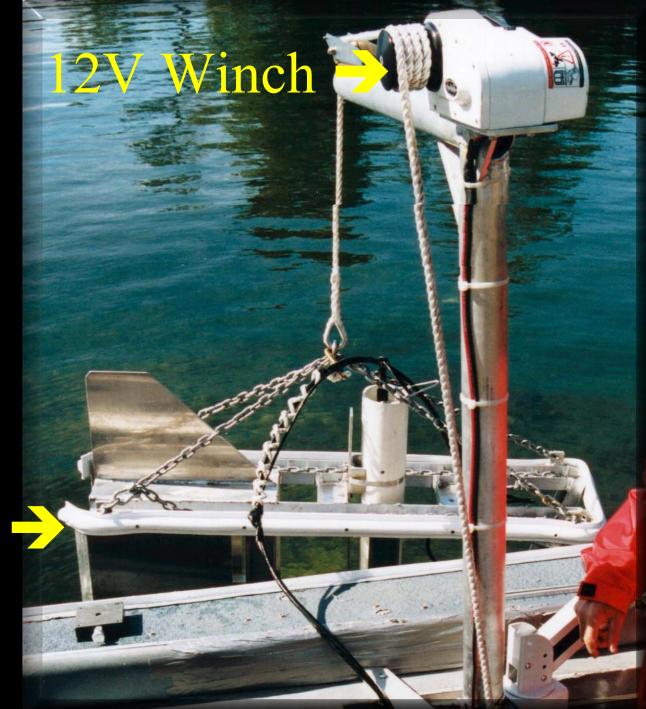




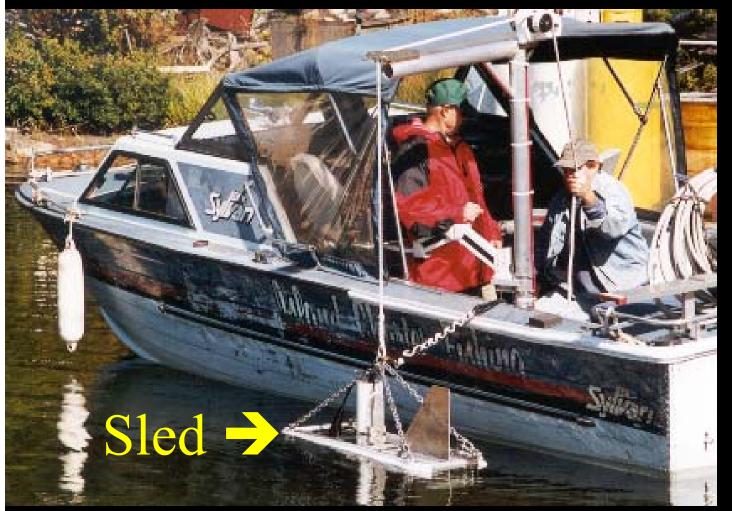


### How do we map?

# Sled ->



# **Instrumentation**



Hydrolab Datasonde



In-situ Mapping Parameters-

 pH, conductivity, dissolved oxygen, Eh, temperature, depth, chloride, turbidity

### Why do we map?

Detect groundwater springs entering lakes and rivers

Groundwater Emergence Sites
•Structural Control

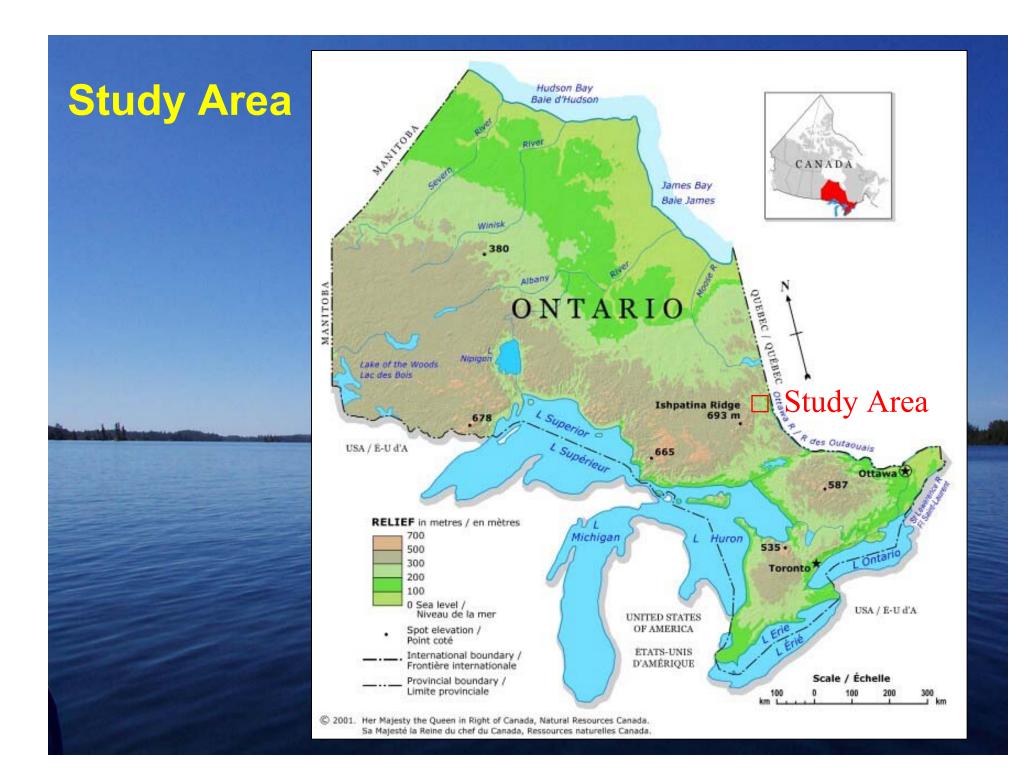
Faults \*

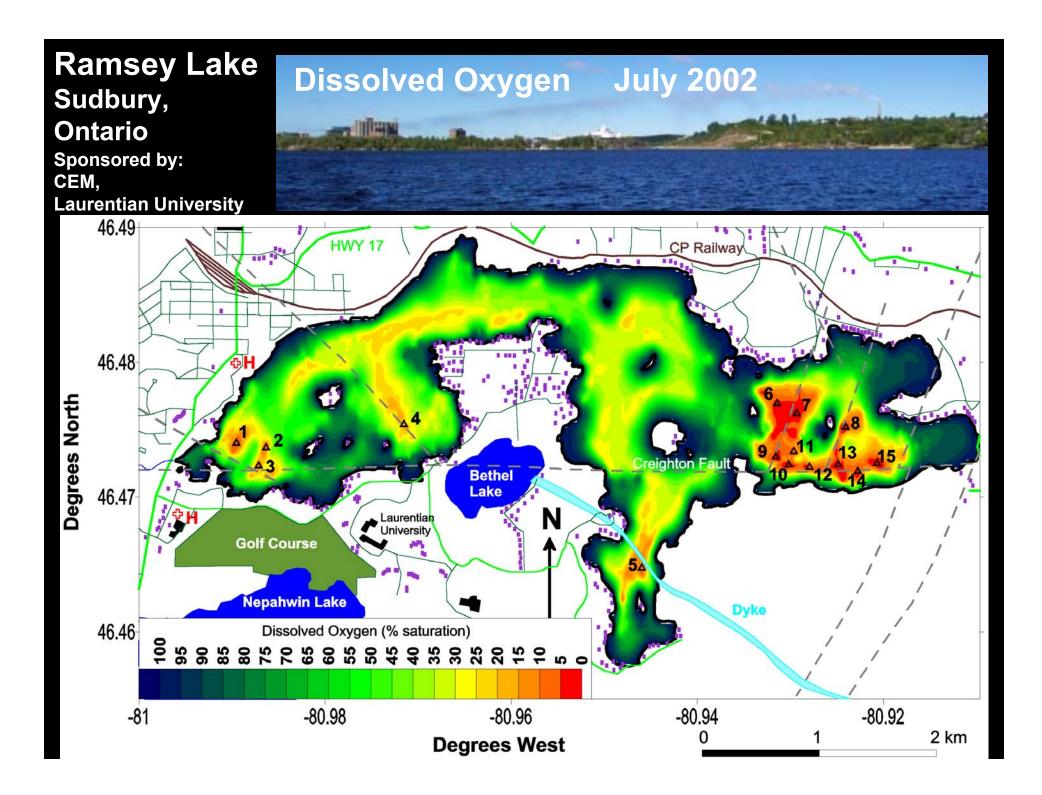
Joints

- Igneous contacts (dikes)
- Stratigraphic boundaries
- Glaciofluvial Conduits

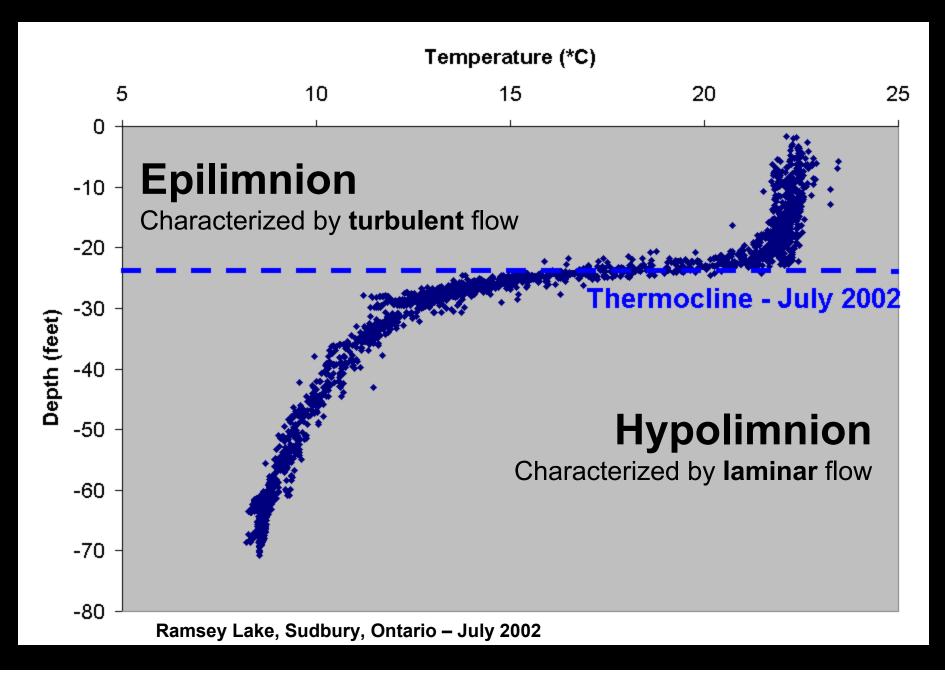
 Relate groundwater chemistry to the bedrock and mineralization with which it has been in contact

 Identify geochemical processes, such as precipitation of sulfides from the groundwater upwelling into surface water





### Where (and when) do we map?



# **Working Definitions**

- **Oxic** =  $DO \ge 15\%$  saturation
- Anoxic = DO < 15% saturation</li>
- Anaerobic = DO < 10% saturation</li>

ORP > 0 mV ORP < 0 mV

Redox potential controlled by anaerobic bacteria producing hydrogen sulfide and/or ammonia

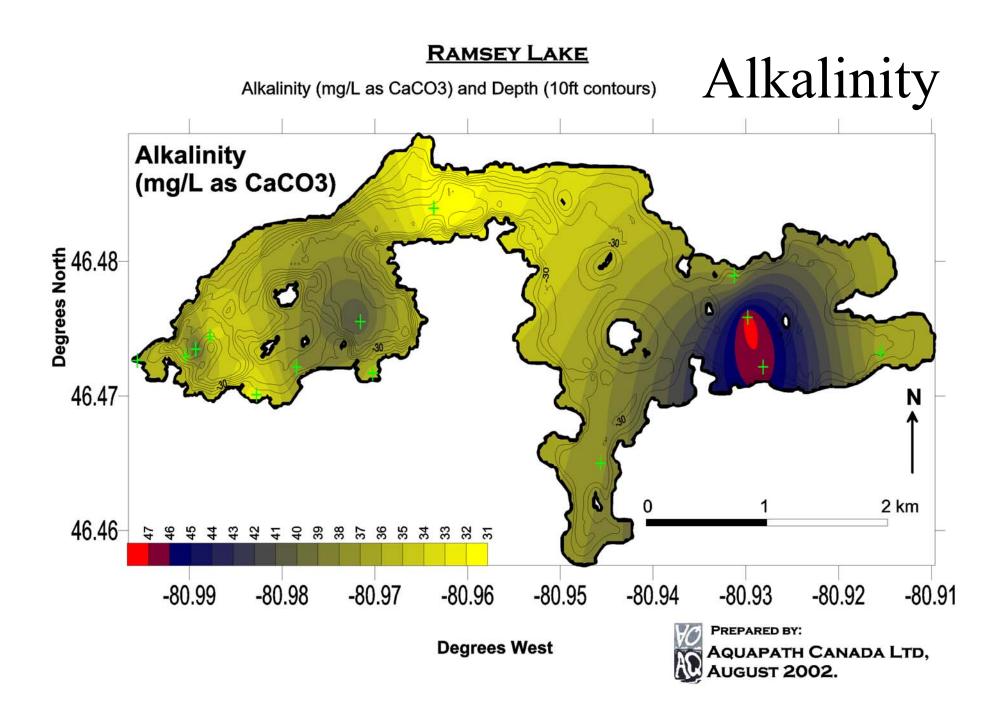
# **Groundwater Characteristics**

Zero % dissolved oxygen

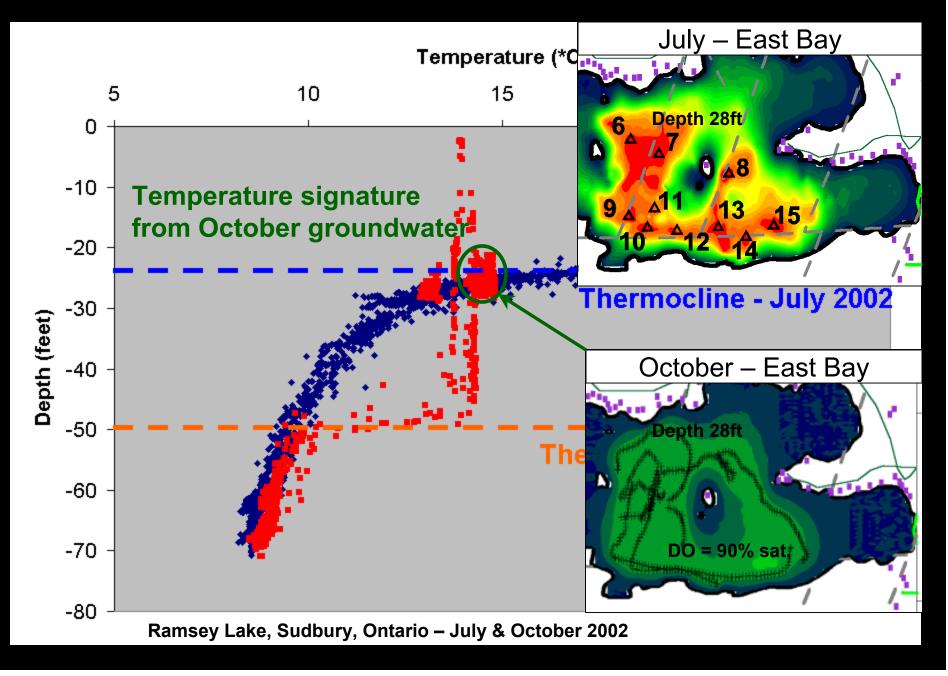
Higher concentration of TDS than lake / river water (usually)

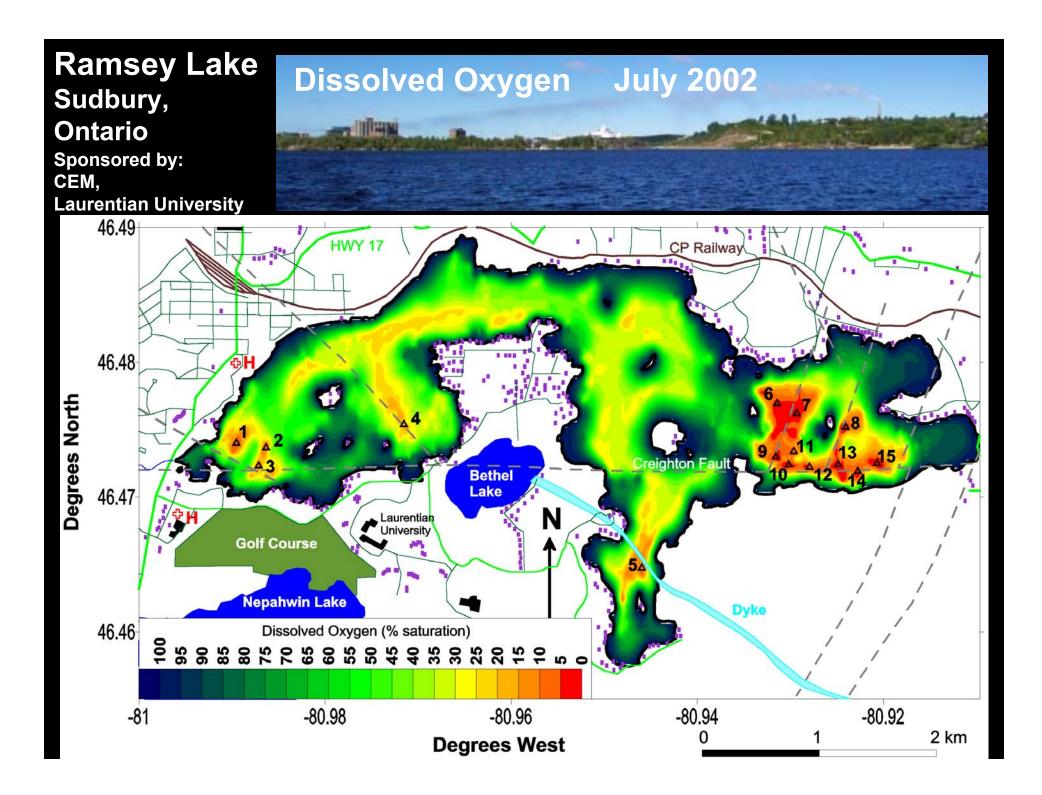
Silica concentration ≥ quartz saturation

• Higher alkalinity (usually)



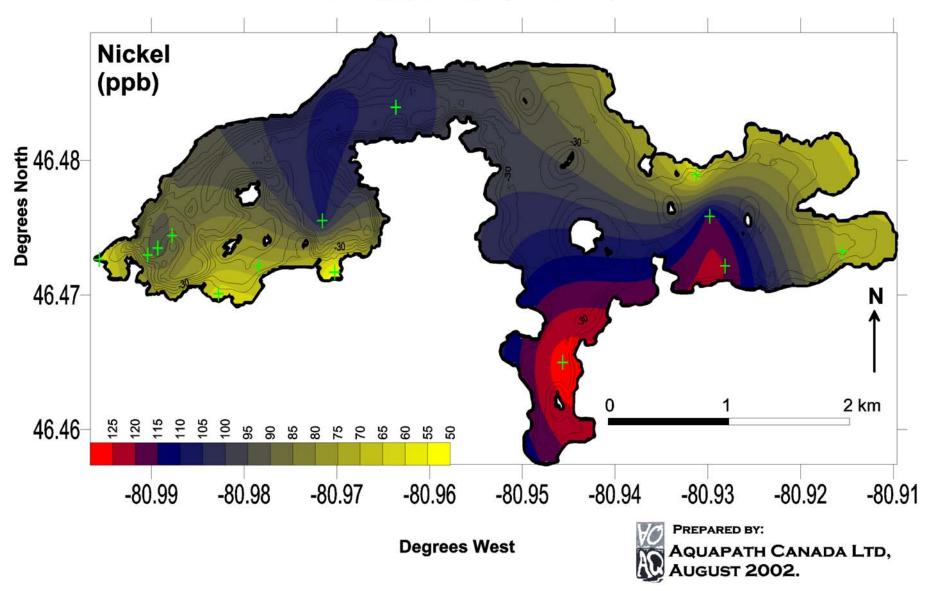
# Where (and when) do we map?





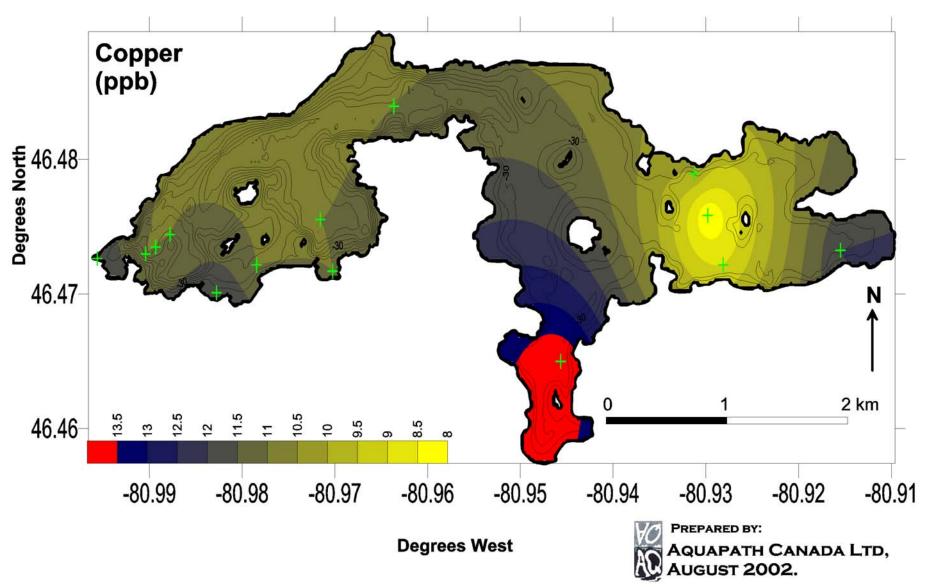
#### RAMSEY LAKE

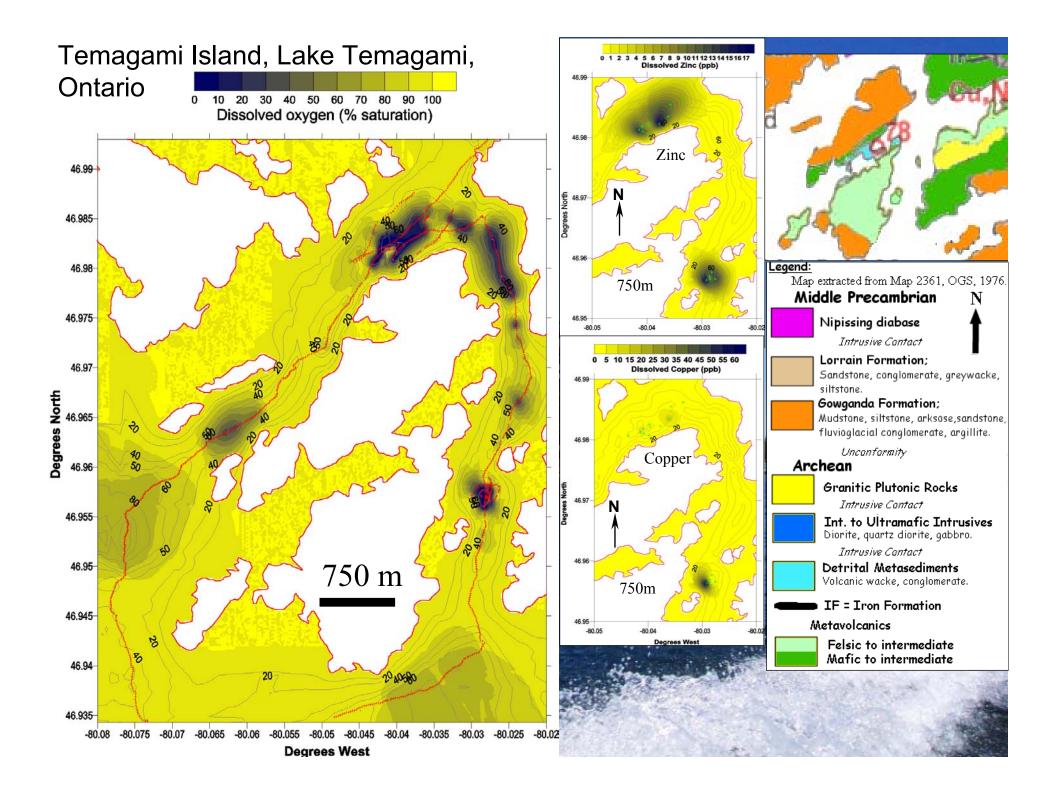
Nickel (ppb) and Depth (10ft contours)



#### RAMSEY LAKE

Copper (ppb) and Depth (10ft contours)

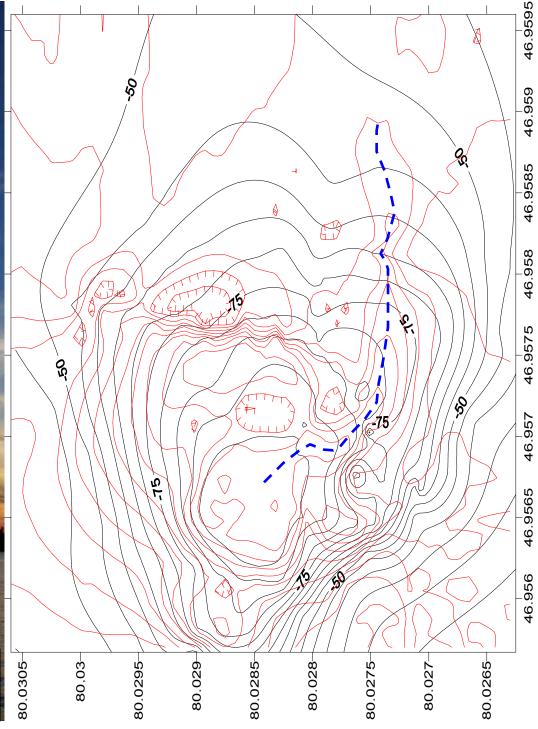






# Strike 082° Dip 4° S





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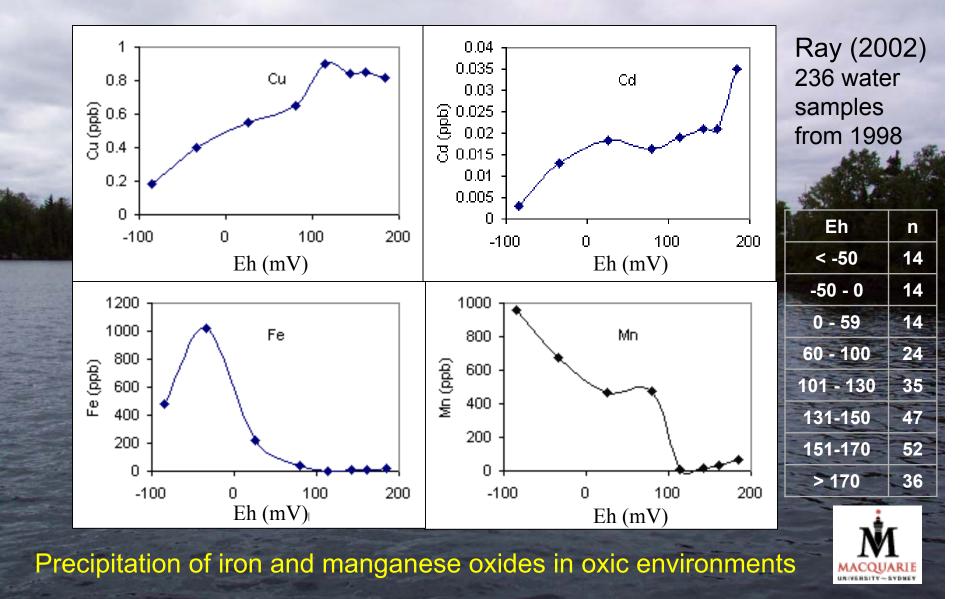
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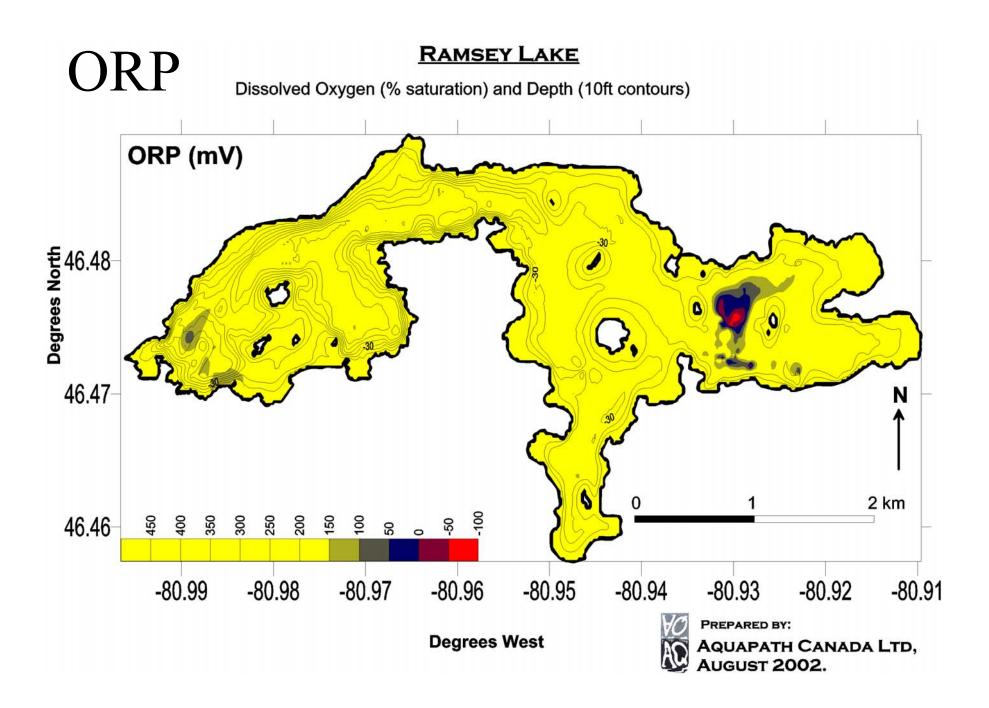
Silica concentration ≥ quartz saturation

• Higher alkalinity (usually)

#### **Distribution of metals under varying Eh conditions**

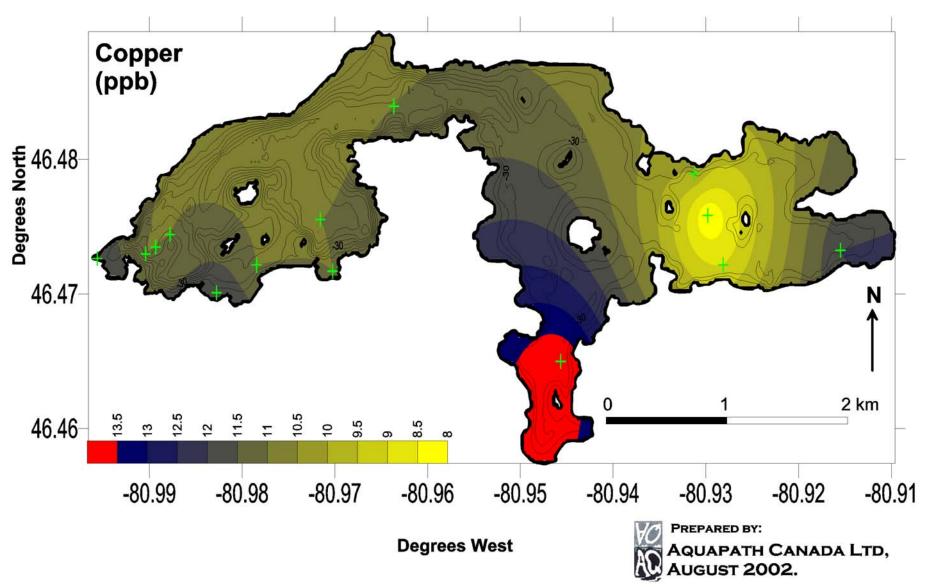
Precipitation of trace metal sulfides such as copper, cadmium & zinc, in some cases depleting soluble levels of these metals below ICP-MS detection limits





#### RAMSEY LAKE

Copper (ppb) and Depth (10ft contours)



### Rates of Groundwater Flow into Lower Circulation Cell

#### **Determinations:**

- Volumes of lower & upper circulation cells
- Time since homogenization
- Mean concentration of upper & lower cells
- Mean concentration of incoming groundwater

#### **Calculations:**

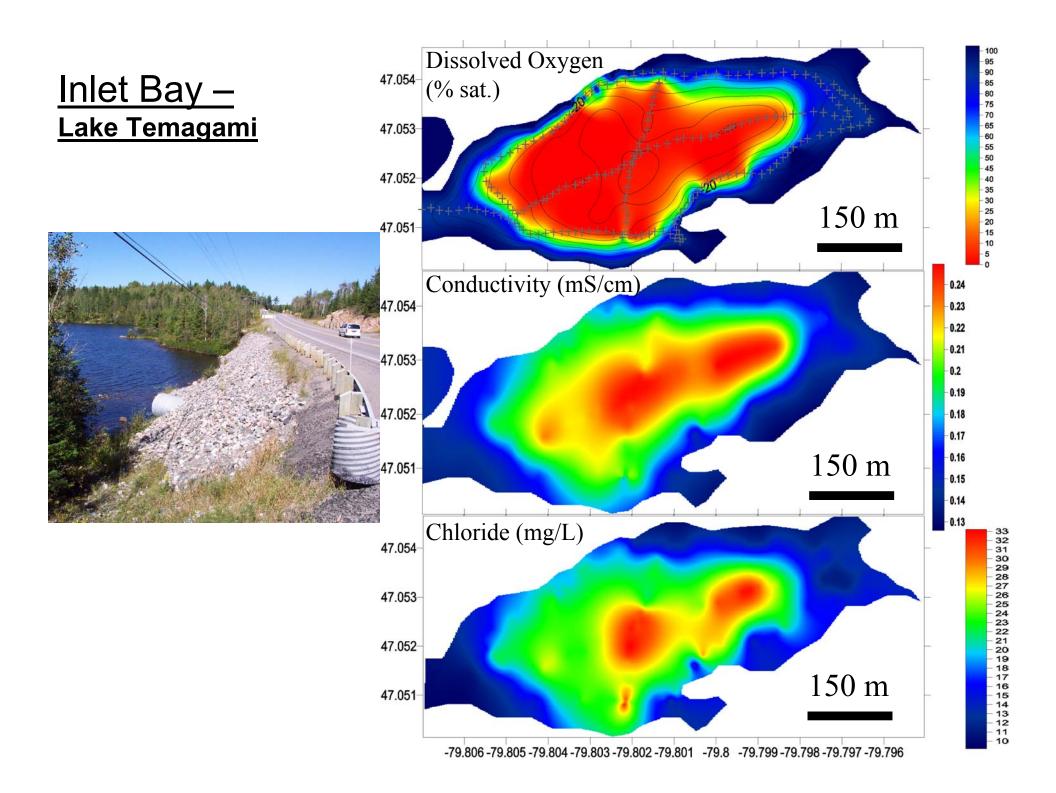
Mean lake water concentration at the time of homogenization (intermediate between present concentration of upper cell and weighted mean of both cells)
Mass of constituent required to reach concentration in lower cell
Litres of groundwater/second to reach this concentration

# **Estimation of Groundwater Flow Rate**

| <b>Constituent</b><br>(nearly conservative) | Groundwater     |
|---|-----------------|
|   | Flow Rate (L/s) |
| Nitrate                                     | 600             |
| Nickel                                      | 580             |
| Fluorine                                    | 480             |
| Phosphorus                                  | 385             |
| TOC   | 270             |

# Removal of Constituents from the Lower Circulation Cell

| Constituent | % remaining since homogenization | te   |
|-------------|----------------------------------|------|
| Nitrate     | 100                              |      |
| Nickel      | 96                               |      |
| Fluorine    | 80                               |      |
| Phosphorus  | 64                               |      |
| Manganese   | 57                               |      |
| Zinc        | 49                               |      |
| TOC         | 45                               | Alle |
| Arsenic     | 14                               |      |
| lron        | 4                                |      |



# Summary

- Aquapath methodology to prepare contour maps of lake and river bottom water in terms of specific sensor variables, such as dissolved oxygen or TDS
- 2. Detect, measure and sample sites of groundwater emergence
- 3. Use groundwater geochemistry to point to source rocks or source material
- 4. Use structural control or glaciofluvial conduit control to point to source material
- 5. Within a lake, or isolated lake basin below the thermocline, determine groundwater input flow rates and removal rates for specific constituents / contaminants

# Acknowledgements

- Aquapath staff
- CEM staff Laurentian University
- OMET and OMET Partners
  - OGS Sudbury
  - OGS Kirkland Lake
  - Activation Laboratories

